

CO₂ uptake in cementcontaining products

Background and calculation models for IPCC implementation

Commissioned by Cementa AB and IVL research foundation

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Preface

The climate issue is today a very important issue both globally in an international perspective and nationally. Many organizations and companies are actively working on climate issues and greenhouse gas reductions are often an important goal, as well as mapping and monitoring of greenhouse gases. At the international level, the different countries report greenhouse gas emissions to the United Nations Framework Convention on Climate Change (UNFCCC). The international monitoring of net greenhouse gas emissions to the atmosphere is an important part of the mapping and monitoring of global climate effects. The guidelines for how emission calculations are to be carried out are presented in the framework of the Intergovernmental Panel on Climate Change (IPCC) and are regulated in, for example, the document "2006 IPCC Guidelines for National Greenhouse Gas Inventories". In order for the calculations of the net greenhouse gas emissions to the atmosphere to be as accurate as possible, it is important that the guidelines and other documents are updated and improved by the IPCC. Consequently, continuous efforts are being made to improve methodology and to update the "2006 IPCC Guidelines for National Greenhouse Gas Inventories".

As part of the climate work for the cement and concrete industry, the estimates of carbon dioxide emissions have been analyzed with regard to the regulations in the "2006 IPCC Guidelines for National Greenhouse Gas Inventories". It shows that CO₂ emissions from cement production are calculated from both the combustion of different fuels in the cement kiln and from the discharging of CO₂ from the raw materials, mainly due to various incoming carbonates. This process is usually referred to as calcination and can be exemplified by the following chemical formula showing the thermal decomposition of limestone:

$$CaCO_3 \rightarrow CaO + CO_2$$

However, the calcination reactions in cement are not chemically stable but are reversible. This means that CO₂ in the air reacts with hydrated cement phases in the concrete and carbonates are regenerated. This process is usually called carbonation. Carbonation can take place during the lifetime of the concrete product, but also in, for example, crushed concrete as a secondary product (e.g. base course of a road or as landfilling material). This uptake of CO₂ in the concrete thus reduces the net emission of CO₂ from the raw material part. This uptake of CO₂ can be significant, which is why it is important to take this effect into account in emission calculations. The current version of the "2006 IPCC Guidelines for National Greenhouse Gas Inventories" does not take into account the effects of carbonation. This could lead to reduced accuracy in both national and global emission calculations.

A significant amount of research is available regarding carbonation (see e.g. Appendix 2). Some CO₂ uptake models have been developed in different countries and a European standard [16] is available, with an annex on calculation of CO₂ uptake in concrete products. However, uptake models adapted to the IPCC applications are missing. This is thus the origin and background of the present study. As calculation models for CO₂ uptake in concrete are difficult and complex, an international research team has been created to capture present research and to further develop appropriate methods and calculation models for CO₂ uptake with the aim of improving net emission estimates of climate gases and providing support for improved calculation methods within the IPCC and especially the "2006 IPCC Guidelines for National Greenhouse Gas Inventories". The study can also be used as a guideline for the national greenhouse gas



calculations. The present report is the result of this research group's work and is intended to be used as a basis for further development work in the climate field.

The present study has been carried out by IVL Swedish Environmental Research Institute as a cofinanced research project. The project is co-financed by IVL research foundation (50 %) and Cementa AB (50 %). IVL's part is part of an international project to develop methods for CO₂ uptake in concrete which has contributed to important scientific information. The results from this study will be provided to IPCC and other organizations in order to include CO₂ uptake in concrete as a part of the global CO₂ net emission calculations.

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Summary

The climate issue is today a very important issue both globally in an international perspective and nationally. Many organizations and companies are actively working on climate issues and greenhouse gas reductions are often an important goal, as well as mapping and monitoring of greenhouse gases. At the international level, the different countries report greenhouse gas emissions to the United Nations Framework Convention on Climate Change (UNFCCC). The international monitoring of net greenhouse gas emissions to the atmosphere is an important part of the mapping and monitoring of global climate effects. The guidelines for how emission calculations are to be carried out are presented in the framework of the Intergovernmental Panel on Climate Change (IPCC) and are regulated in, for example, the document "2006 IPCC Guidelines for National Greenhouse Gas Inventories". In order for the calculations of the net greenhouse gas emissions to the atmosphere to be as accurate as possible, it is important that the guidelines and other documents are updated and improved by the IPCC. Consequently, continuous efforts are being made to improve methodology and to update the "2006 IPCC Guidelines for National Greenhouse Gas Inventories".

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Inventories". The study can also be used as a guideline for the national greenhouse gas calculations.

In principle, the same amount of CO_2 driven off in the cement kiln can be taken up in the concrete by carbonation. However, the amount of CO_2 that will be taken up by carbonation in reality depends on several factors. The carbonation process is a slow process that can last for many years. The time aspect is thus an important issue. The availability of CO_2 for the concrete is also crucial. The concrete must be exposed to the CO_2 in the air to be able to carbonate. The transport of CO_2 molecules into the concrete is thus also an important factor as well as moisture factors. For example, if the concrete is crushed after use, the carbonation rate will increase considerably due to increased concrete surface areas.

Carbonation is thus an important aspect to take into account in climate and emission calculations for cement and concrete. Today, emissions of greenhouse gases from the different countries are reported, which in turn are used to support different climate strategies. Reporting takes place nationally to national authorities and internationally to the United Nations Framework Convention on Climate Change (UNFCCC). Both CO₂ emissions from fossil fuel combustion and emissions from the raw materials (calcination) are included. However, no consideration is given to the carbonation of concrete. This may be considered a shortcoming in these calculations, which can lead to less accurate results. A very rough estimate is that the use of concrete today accounts for about 5 % of the world's carbon dioxide emissions. About half of these emissions comes from the raw materials and thus has a potential to be reabsorbed by carbonation of the concrete, partly during the use phase of the concrete products, and partly in the end-of-life and secondary use stage. This study proposes methods and models for calculating CO₂ carbonation in various cement-containing products, including mortar for rendering and cement/concrete additions such as ground granulated blast-furnace slag (GGBS), fly ash, and lime filler.

Model calculations of national CO₂ uptake in concrete are rather complex and involve many different factors that control the uptake. However, the calculations can also be simplified but with an increased uncertainty. In-line with the emission calculations developed for the "2006 IPCC Guidelines for National Greenhouse Gas Inventories", three different calculations methods are developed in this study, Tier 1, Tier 2 and Tier 3, in increased model complexity. According to IPCC rules, Tier 2 and 3 should replace Tier 1 if the uptake in the country is defined as a "key category" either on "level" or "trend". The calculation methods are based on previous research in the area. Tier 1 is a simplified calculation method that can be used by almost all national calculation groups with a minimum of input data. The proposed Tier 1 average value is 0.23 (23 %) of the calcination from the process. This includes the annual CO₂ uptake in existing concrete structures, in end-of-use and secondary use. Tier 2 is a more complex calculation method that can be used in the national calculation groups. Tier 3 opens up for the use of complex computer models for the uptake calculations that requires more knowledge and input data. Such models have been developed in a few countries on a research basis.



Abbreviations and definitions

Calcination: Is, in this case, defined as the thermal decomposition of mainly calcium carbonate

(CaCO₃) or other carbonates under the formation of CO₂.

Carbonated concrete: Part of concrete where carbonation has occurred and colorless indication

with phenolphthalein solution shows that pH has been lowered to less than about 8. Additional CO_2 uptake may have occurred in areas where pH

change has not yet taken place.

Carbonated zone: Part of concrete that is carbonated.

Carbonation depth: Distance from surface to where colorless indication with phenolphthalein

solution shows that pH has been lowered to less than about 8. Additional uptake of CO₂ may have occurred beyond the carbonation front, prior to pH

change.

Carbonation rate: The carbonation rate depends of several factors such as the chemical reaction

rate, mass transport of CO₂, humidity, temperature, porosity, CO₂ concentration in ambient air etc. For practical reasons the carbonation rate is often determined by measuring the depth of carbonation as function of time.

It can then be expressed as mm/ $\sqrt{\text{year}}$.

Carbonation: A chemical reaction by which CO₂ penetrates the concrete and reacts with the

hydration products, forming mainly calcium carbonate.

CEM I: Is a pure Portland cement.

CEM II: Portland composite cement containing at least 65 % Portland clinker.

CEM III: Contains at least 20 % and at most 65 % Portland cement and the remaining is blast-

furnace slag.

Cement: Binder in concrete, and mortar and made of ground clinker, gypsum and often some

added constituents such as GGBS, fly ash, silica fume, or limestone.

Clinker: Is the sintered material produced in the cement kiln when heating the raw mix to high

temperature.

CO₂ uptake area: Is the area in cement containing products where CO₂ can be absorbed through

carbonation.

Concrete: Construction material made mainly of cement, gravel, sand, water, and possibly of

additions and admixtures.

Degree of carbonation: The amount of CO₂ uptake by carbonation in relation to maximum CO₂

uptake by carbonation. The maximum uptake of CO_2 can often be equated with the CO_2 emission from calcination that is driven off from the material. Traditionally, the degree of carbonation has been defined only within the area which has been considered carbonated and this area has been defined as the area exhibiting color change with

phenolphthalein test.

DOC: Degree of carbonation.

GGBS: Ground granulated blast-furnace slag.



Gt C: Gigaton carbon.

IPCC: Intergovernmental Panel on Climate Change.

Mortar: Is a mix of cement, sand and water. It is used as bonding agent between building

materials or for rendering of surfaces.

MR: Mortar used for rendering surfaces.

Portland cement: Is a common type of cement made of mainly grounded clinker and 2-3 %

gypsum.

Ton: metric ton = 1000 kg

UNFCCC: United Nations Framework Convention on Climate Change.

z factor: z-score percentile for normal distribution.



1 Introduction

1.1 Background

Concrete is the single most important building material in society and is used for a variety of products, such as houses, bridges, tunnels, roads, roof tiles, and other major and minor construction products, etc. These products constitute the primary use of concrete. After the lifetime of these products, the concrete is normally crushed and used as secondary products e.g. in roads or base course, or as fill material. The binder in concrete consists of cement. The most common cement type, Portland cement, is made by mixing limestone with other materials such as iron, aluminum and silicon containing minerals, usually in the form of clay. The materials are ground, mixed and burned at high temperature (1400-1450°C) in a rotary kiln to form cement clinker. In this process, the materials sinter to form clinker. Other materials may also be present in cement manufacture such as blast-furnace slag and fly ash. These materials are examples of the use of recycled materials in the manufacture.

The production of clinker requires high temperature and is therefore very energy intensive. The fuels used globally are mainly coal, oil and pet coke (from oil refining), but also fuels made from residues such as waste oil, solvents, plastic and waste tires are used. The residual products can, in this way, be treated safely and efficiently in high temperature, long residence time, and with extensive cleaning of the exhaust gases while the energy content can be used. The combusted residues can, in this way, replace and reduce the use of fossil fuels. The fossil fuels used in the cement kiln give rise to an emission of CO_2 that will contribute to the global warming. Emissions of other greenhouse gases such as methane (CH_4) and nitrous oxide (N_2O) are relatively small since the combustion temperature is high and the combustion condition is well-controlled in the cement kilns.

In the production of cement, most of the carbon dioxide is formed partly from the combustion of the fuels needed in the production (cement kiln) and partly from the calcination of the limestone, e.g. according to the reaction below:

$$CaCO_3 \rightarrow CaO + CO_2$$

These calcination reactions are not permanent but reversible. This means that CO_2 is absorbed into the concrete by a process referred to as carbonation. In principle, the same amount of CO_2 driven off in the cement kiln can be taken up in the concrete by carbonation. However, the amount of CO_2 that will be taken up by carbonation in reality depends on several factors. The carbonation process is a slow process that can last for many years. The time aspect is thus an important issue. The availability of CO_2 for the concrete is also crucial. The concrete must be exposed to the CO_2 in the air to be able to carbonate. The transport of CO_2 molecules into the concrete is thus also an important factor. For example, if the concrete is crushed after use, the carbonation rate will increase considerably.

Carbonation is thus an important aspect to take into account in climate and emission calculations for cement and concrete. Today, emissions of greenhouse gases from the different countries are reported, which in turn are used to support different climate strategies. Reporting takes place nationally to national authorities and internationally to the United Nations Framework Convention on Climate Change (UNFCCC). Guidelines for the emission calculation are developed and kept



updated by the Intergovernmental Panel on Climate Change (IPCC). The latest version, "2006 IPCC Guidelines for National Greenhouse Gas Inventories," covers greenhouse gas emissions from cement and concrete processes. Both CO₂ emissions from fossil fuel combustion and emissions from the raw materials (calcination) are included. However, no consideration is given to the carbonation of concrete. This may be considered a shortcoming in these calculations, which can lead to less accurate results. A very rough estimate is that the use of concrete today accounts for about 5 % of the world's carbon dioxide emissions. About half of these emissions comes from the raw materials and thus has a potential to be reabsorbed by carbonation of the concrete, partly during the use phase of the concrete products, and partly in the end-of-life and secondary use stage. We are convinced that this is an important part of the climate calculations and we therefore want to improve the calculations to better reflect the reality. This study proposes methods and models for calculating CO₂ carbonation in various cement-containing products.

1.2 Purpose and goal

The main purpose of the study is to provide input to the national and international greenhouse gas inventories by developing new calculation models for CO₂ in the cement and concrete sector. The goal is to include uptake of CO₂ by carbonation in concrete both during the lifespan of the concrete products and during their end-of-life and secondary use. Accurate measurements and good data are always important for all decision-making processes and this also applies to the climate issue. The calculation methods to be used to calculate the uptake of CO₂ in concrete have, to some extent, been based on previously performed scientific work and models, but new methodology for uptake calculations has also been proposed in this study. The study includes preparation of calculation methods with different complexity and accuracy (3 methods). In the IPCC's guidelines, such a level differentiation is often referred to as Tier 1, Tier 2 and Tier 3, which has also been used in this study. The method that a country will use depends on the country's ability to access statistics on cement/concrete and its use in different products. The methods used will also depend on if it, in the country, will be defined as a "key category" according to IPCC. The uptake of CO₂ in concrete, both in primary products and in end-of-life & secondary product use, is relatively slow processes that also depend on the exposure of CO₂ from the air. The CO₂ content in air today is about 400 ppm and increases by about 1-2 ppm per year. However, almost all existing concrete takes up CO₂ continuously, while the production of cement continuously emits CO2. This results in a kind of annual net emission of CO₂. It is this net emission that is the focus of the calculations although the time aspect itself also is addressed in the project. The different calculation methods relate to an annual uptake but since they are time-generic they can also be used to develop different time series.

Since CO₂ uptake in cement-containing products affects several percentages of the global CO₂ emissions, this represents a significant aspect. In addition, the issue concerns several major industrial sectors. It mainly concerns the cement and concrete sector, but the construction sector and infrastructure construction sector can also be affected. For each industry sector, it is of course of great importance that data from the sector is accurate and that the sector's activities can be judged properly.

Publishing this knowledge also give new insight in how to further increase the annual CO₂-uptake.



2 Theoretic background to CO₂ uptake in cement-containing products

The chemistry of cement and concrete include a large amount of chemical reactions to achieve the specific property of the concrete product. Different raw materials and additives are also used that can influence the chemical reaction. The focus in this study is the climate performance and behavior of different concrete products and especially the calculation of its carbon dioxide (CO₂) balance. In this chapter, we will thus present the fundamental chemical reactions of CO₂ and its corresponding flows. A more detailed description of the processes can be found in, for example reference [11].

Portland cement is made mainly of four different components; calcium, silicone, aluminum and iron. These are supplied in cement production through various raw materials such as limestone, clay, marl, silica sand, shale, etc. (the raw mix). A main raw material in cement is limestone (CaCO₃). In the cement kiln, the raw materials are heated up and CO₂ is driven off in the calcination reaction mainly from CaCO₃ according to reaction (1). In the calcination reaction, lime (CaO) and CO₂ is formed. CO₂ is released to the atmosphere in the exhaust gases from the cement kiln. CaO reacts further to form an integral part of the cement mainly as various calcium silicates, calcium aluminates, and calcium ferrites. Only a small part occurs as free CaO.

$$CaCO_3 + heat \rightarrow CaO + CO_2$$
 (1)

In the manufacture of concrete, water is added to the cement to form the cement paste (hydration process). The added water will then react with different substances in the cement such as tricalcium silicates and dicalcium silicates to form hydration products such as calcium silicate hydrates (C-S-H) gel and also Ca(OH)₂. A comprehensive presentation of cement reactions can be found in for instance [18]. The carbonation reaction is often written, for simplicity, with Ca(OH)₂ but carbonation will occur also in the other components, such as the C-S-H gel.

Thus, both C-S-H gel and $Ca(OH)_2$ form a part of the cured concrete. CO_2 in the atmosphere, in contact with concrete, will primarily react with $Ca(OH)_2$ in the concrete according to the principle reaction (2) but will also react with the C-S-H gel. These reactions represent the uptake of CO_2 in concrete, which is called carbonation.

$$Ca(OH)_2 + CO_2 \rightarrow CaCO_3 + H_2O$$
 (2)

 CO_2 is a natural part of the atmosphere. However, the concentration of CO_2 in the atmosphere is increasing due to an extensive global use of fossil fuels. The concentration of CO_2 in the atmosphere has increased from about 280 ppm in preindustrial time to about 400 ppm today. The rate of concentration increase in the atmosphere is today about 1-2 ppm/year. An increased concentration of CO_2 in the atmosphere can, to some extent, also increase the rate of carbonation.



The carbonation reaction takes place in several steps. The actual uptake reaction is the reaction between the calcium and carbonate ions (3). This reaction takes place in water phase in the pore solution in the concrete. Water and moisture is thus an important part of carbonation.

$$Ca^{2+} + CO_3^{2-} \rightarrow CaCO_3 \tag{3}$$

In the alkaline pore water solution in concrete, portlandit (Ca(OH)₂) can be dissolved according to reaction (4) forming calcium and hydroxide ions.

$$Ca(OH)_2 \rightarrow Ca^{2+} + 2OH^{-} \tag{4}$$

 CO_2 is also dissolved in the alkaline pore water solution according to reaction (5) forming carbonic acid (H_2CO_3).

$$CO_2 + H_2O \rightarrow H_2CO_3 \tag{5}$$

The protolysis of H_2CO_3 in alkaline solution proceed in two steps (6) and (7) forming bicarbonate (HCO_3^-) and carbonate (CO_3^{-2}) ions.

$$H_2CO_3 + OH^- \leftrightarrow HCO_3^- + H_2O$$
 (6)
 $HCO_3^- + OH^- \leftrightarrow CO_3^{-2-} + H_2O$ (at high pH, in uncarbonated concrete) (7)

In this way, Ca^{2+} and CO_3^{2-} are formed and can react and precipitate as limestone (CaCO₃) in the concrete

An important aspect in carbonation is the carbonation rate. How fast can the carbonation proceed and which are the rate determining processes? When estimating the CO_2 uptake in the carbonated concrete one must also consider the degree of carbonation.

The *carbonation rate* depends on several factors such as the chemical reaction rate, mass transport of CO_2 , humidity, temperature, porosity, CO_2 concentration in ambient air etc. The rate determining step could be crucial to the overall uptake rate. Of practical reasons, the carbonation rate is often measured by measuring the depth of carbonation as a function of time. The carbonation rate can then be expressed as mm/ \sqrt{y} ear.

The degree of carbonation is defined as the amount of CO_2 uptake by carbonation in relation to maximum CO_2 uptake by carbonation. The maximum uptake of CO_2 can often be equated with the CO_2 emission from calcination that is driven off from the material. Traditionally, the degree of carbonation has been defined only within the area which has been considered carbonated and this area has been defined as the area exhibiting color change with phenolphthalein test. However, when calculating CO_2 uptake in concrete or other cement-containing products, there may be a need for a more general definition of degree of carbonation that includes all carbonated CO_2 in relation to all concrete.

In general, one can say that the chemical reactions presented above are relatively fast and cannot be considered as the rate determining step. However, there are several other factors that can slow down the carbonation rate. A common aspect for those factors is the access to and transport of molecules in the concrete. As carbonation proceeds, more and more CaCO₃ is precipitated in the concrete and can thus also cover Ca(OH)₂ particles, or reduce the permeability of the concrete. This



will slow down the dissolution of $Ca(OH)_2$ and reduce the access of CO_2 to the interior of the concrete, and thus decrease the carbonation rate.

As also has been mentioned, water is required for the carbonation to take place. Concrete is a porous material that allows both CO_2 in air and water to penetrate into the concrete. The CO_2 gas in the pores will dissolve in the water in the pores and carbonation can start. However, if the pores are completely filled with water, the HCO_3^- or CO_3^{2-} ions have to diffuse in the water phase into the concrete. This is a much slower process and will thus slow down the carbonation rate. Obviously, there is an optimal moisture content in concrete for a maximum carbonation rate. The optimal moisture content in concrete for carbonation has been estimated to be about 60-80 % relative humidity in the concrete.

Mainly due to the formation of $CaCO_3$ in the concrete, the carbonation rate will slow down with time. Empirical experiments have shown that the carbonation rate is proportional to the square-root of time (t), \sqrt{t} [11,19]. Other factors that will influence the carbonation rate are porosity of the concrete, w/c ratio, cracks in the concrete, cement type and additives, and surface treatment of the concrete products. At the end-of-life of concrete products, they are demolished and sometimes crushed to recover the iron reinforcement bars. When the concrete is crushed, new surfaces are created and smaller concrete pieces are formed. This can dramatically increase the carbonation rate if access to CO_2 in air can be maintained. Smaller concrete pieces will also increase the total carbonation in an entire concrete volume. To estimate the uptake of CO_2 in concrete, it is thus important to include both the service life of the concrete products and the secondary use of the concrete after the end-of-life phase.



3 IPCC documentation of CO₂ emissions and uptake

Annual national greenhouse gas (GHG) emissions and removals are reported by countries to international databases such as UNFCCC through their operations "Greenhouse Gas Inventory". The UNFCCC formulates its own activities as follows:

"The ultimate objective of the Climate Change Convention (UNFCCC) is to achieve "... stabilization of greenhouse gas concentrations in the atmosphere at a level that would prevent dangerous anthropogenic interference with the climate system." Estimating the levels of greenhouse gas (GHG) emissions and removals is an important element of the efforts to achieve this objective."

These databases can then be used to help assess the development of the climate from a global perspective and to develop global and regional climate strategies. It is therefore important that the uptake of CO₂ by carbonation in concrete and other cement-based products is included in the international emissions reporting so that accurate data on the net supply of CO₂ to the atmosphere through the use of cement and concrete are obtained.

Calculation of greenhouse gas emissions for reporting to the UNFCCC is made by nationally appointed organizations in each reporting country. Guidelines for how calculations of the climate gases are to be made are determined by control documents developed by the IPCC. The control documents can be found on the IPCC website: https://www.ipcc-nggip.iges.or.jp/public/2006gl/

The main document that regulates emission calculations and removals is 2006 IPCC Guidelines for National Greenhouse Gas Inventories. The guideline consists of five volumes:

- Volume 1 General Guidance and Reporting
- ➤ Volume 2 Energy
- Volume 3 Industrial Processes and Product Use
- Volume 4 Agriculture, Forestry and Other Land Use
- Volume 5 Waste

The applicable emission calculations are divided into emissions from energy production and emissions from processes. The emissions from the cement kiln can thus be divided into emissions from the combustion and CO_2 emissions emanating from the raw material in the calcination process. The latter CO_2 emission is classified as a process emission and regulated in Volume 3 Industrial Processes and Product Use. The CO_2 uptake in concrete is strongly related to the CO_2 emissions from the raw material and it is thus proposed that the CO_2 uptake will be regulated in the same document (Volume 3) as the process emissions. This volume of the IPCC guidelines also includes emission and uptake in product use, which is the case for CO_2 uptake in concrete products. Product use can here also include CO_2 uptake in end-of-life and secondary products i.e. for example in crushed concrete.

Volume 3 Industrial Processes and Product Use includes several industrial processes of which Chapter 2 Mineral Industry Emissions is the most appropriate for this task. CO₂ uptake can occur in several of these product systems, which is why it is important to prepare the methodology for additional industrial processes. The uptake chapter can also be separated from the emission



chapters. A revised chapter structure of Volume 3, Chapter 2 can thus be as follows with changes marked in bold:

Chapter 2 Mineral Industry Emissions and removals

- 2.1 Introduction
- 2.2 Cement production
- 2.3 Lime production
- 2.4 Glass production
- 2.5 Other process uses of carbonates
- 2.6 CO2 removals
 - 2.6.1 CO₂ uptake in cement containing products

The present study can thus be an input to a revision of the 2006 IPCC Guidelines for National Greenhouse Gas Inventories but it can also serve as a calculation guideline for the nationally appointed emission calculation organizations as well as input to IPCC's emission factor database (EFDB).



4 General description of CO₂ uptake models for cement-containing products

As shown in Chapter 3, the CO₂ emission model for the raw material to the cement kiln is presented in the document: 2006 IPCC Guidelines for National Greenhouse Gas Inventories. The calculated CO₂ emission is based on the amount of cement clinker from the cement kiln. This calculated amount of CO₂ driven off from the material can be considered as the maximum theoretic uptake of CO₂ due to carbonation of the different cement containing products. The CO₂ uptake model will thus calculate the CO₂ uptake in the different cement containing products during their lifetime as well as in the end-of-life processes and when used as secondary products such as crushed concrete in a road base or as landfilling material. The emission and uptake models are illustrated in Figure 1.

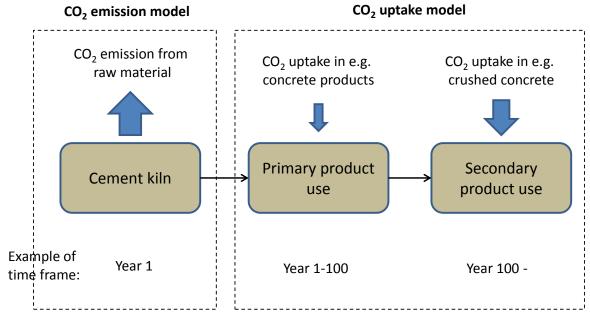


Figure 1 Schematic figure showing the CO₂ balance in cement containing products over a certain period of time.

The CO₂ uptake in e.g. concrete is a relatively slow process that takes place during many years. The first uptake phase is uptake in e.g. concrete products such as bridges, house frames, concrete tiles, concrete roads, railway sleepers, cement mortar etc. The carbonation process takes place from the surface of the concrete product when CO₂ in air diffuse into the porous concrete and reacts with Ca(OH)₂ in the pore solution to form CaCO₃ according to the reactions described in Chapter 2. The area of concrete surfaces or the area/volume ratio of concrete products are important factors for a CO₂ uptake model. By knowing the total yearly use of cement clinker and its use in different product categories, the CO₂ uptake areas can be estimated. The yearly use of cement clinker in a country can be calculated as (cement clinker production-clinker export+clinker import). From the



uptake areas, the yearly CO_2 uptake over the years can be calculated. For large concrete product, only a smaller part of the concrete will be carbonated during its primary product use. Another factor that influences the CO_2 uptake in the concrete surfaces is the moisture content, see Chapter 2. The surface can be located in different climates, be exposed to rain or located indoors/outdoors. These factors can affect both the carbonation rate and the degree of carbonation. Also hydraulic concrete additives, such as blast-furnace slag or fly ash from coal combustion, which also can take up CO_2 , can be important to include in the calculations.

After the service life of a concrete structure, it will be demolished and eventually crushed into finer pieces. This will increase the specific CO₂ uptake area and increase the carbonation rate. The total carbonation in the entire concrete volume will also be increased when the concrete is crushed into smaller pieces. However, the use of the crushed concrete must be done in such a way that air and CO₂ are allowed to access the increased concrete and CO₂ uptake surfaces. This may require some active planning of the end-of-life/secondary use processes for concrete. The knowledge about the end-of-life/secondary use processes for concrete in different countries is lower and the uncertainties are greater. The CO₂ uptake models for the secondary product use are therefore relatively uncertain. However, the CO₂ uptake potential is generally large for the end-of-life/secondary use phase.

In a CO₂ uptake model, many different details can be included to improve the model performance. However, there is also a need to make simplified model calculations based on a few input data. A method to achieve this is to; only use the use of cement clinker for the model calculations. The maximum CO₂ uptake in the clinker used can be multiplied by a factor to compensate for the real uptake in both the primary use and end-of-life/secondary use phase. When using the present year's cement clinker production to calculate the uptake, one has to note that; the CO₂ uptake actually takes place in the previous year's concrete production. If the cement clinker use varies considerably over the years, this may cause model deviations that may need to be compensated for in the models. This is further discussed in Chapter 5.



5 Proposed CO₂ uptake models for cement-containing products

5.1 Overview description and strategies

This chapter presents the calculation methods for the uptake of CO₂ in cement-containing products developed in this study. Here is also the basis for the calculation methods presented. Three different calculation methods and levels have been developed, which have different calculation complexity and accuracy. Tier 1 represents a general but simplified calculation method for the uptake of CO₂. Tier 2 and 3 represent more accurate but complex calculation methods, which are preferred if sufficiently good input data on the use of cement in concrete applications are available. Tier 2 is a proposed advanced methodology including several aspects that will affect the CO₂ uptake. Tier 3 opens up for the use of even more advanced and accurate methods and models developed in scientific projects in different countries.

The different calculation methods relate to an annual uptake but since they are time-generic they can also be used to develop different time series.

5.2 Tier 1 CO₂ uptake model - Simplified methodology

5.2.1 Proposed CO₂ uptake model for Tier 1

The present model, representing Tier 1, provides a simplified calculation method for estimating the annual uptake of CO₂ in existing concrete structures on a national basis. The model should be used primarily in cases where resources are missing to perform more accurate calculations according to the calculation methods described for Tier 2 and Tier 3 of Chapter 5.3 and 5.4, since the uncertainty is relatively high for Tier 1 as this is a simplified calculation method. The national annual CO₂ uptake in concrete; in use stage (existing structures), in end-of-use stage (demolition, crushing, stockpiling), and in secondary use, can be estimated according to this Simplified methodology. The uptake values are related to the reported calcination emissions from the consumed clinker (produced-export+import) in the corresponding country. Note that the nationally calculated emission values may need to be adjusted for exports and imports of cement/clinker.

Use stage

Two alternative calculations, A and B, for the CO₂ uptake in the use stage are given. The reason for giving two values is to give options how to handle the uncertainty of the calculated figures. From Chapter 5.2.3 it follows that one can present a calculated a mean value (A) or a low (=Mean - St. dev.) value (B). The IPCC recommendation is to not overestimate, nor underestimate in the calculations, which should indicate alternative A. Uncertainties are there handled by moving to higher Tier-levels.



Alternative A:

The annual uptake in the <u>use stage</u> can be estimated as 0.20×(the reported emission from calcination of consumed cement clinker).

If the mortar for rendering applications, in total, amount to more than 10 % but less than 30 % of the cement consumption, the annual uptake factor in the <u>use stage</u> can be estimated at 0.20 + 0.0115(MR - 10), where MR is the mortar percentage for rendering.

Alternative B:

The annual uptake in the <u>use stage</u> can be estimated as 0.15×(the reported emission from calcination of consumed cement clinker).

If the mortar for rendering applications, in total, amount to more than 10 % but less than 30 % of the cement consumption, the annual uptake factor in the <u>use stage</u> can be estimated at 0.15 + 0.01(MR - 10), where MR is the mortar percentage for rendering.

End-of-life stage and secondary use

Annual uptake in the end-of-life stage and secondary use can be estimated at $(0.02 + 0.01) \times ($ the reported emission from calcination of consumed cement clinker)

Alternatively, the following estimation can be done in the end-of-life stage and the secondary use.

- If the annual amount of concrete being taken out of service and processed on a recycling plant is known, the CO₂ uptake in the end-of-life stage can be calculated to 10 kg CO₂/m³ concrete.
- If the annual amount of crushed concrete, entering the <u>secondary use</u> as unbound material, is known, the uptake can be calculated to 10 kg CO₂/m³ concrete.

The following chapters explain the background and consideration that underlie the proposed model.

5.2.2 Introduction

CO₂ uptake takes place in cement-based concrete or mortar and not in the unhydrated cement powder, but there is of course a relation between the amount of cement used and the amount of concrete produced. This relationship can be used for the design of a simplified methodology for CO₂ uptake in concrete. It can be assumed that the CO₂ uptake one year in a given region or country is related to the cement consumption in the same area. Since the cement consumption normally does not vary much in a few years perspective, it can be feasible to take the year of the reported CO₂ emission from the material as basis for the uptake. One strong motivation for this is also the fact that, due to the square root of time dependency of the uptake, the first years of carbonation and CO₂ uptake of a structure are most important. (During the first 5 years, 22 % of the 100-years uptake takes place and 50 % of the 20-years uptake.)

The process-related CO₂ emission at cement production is almost solely coming from the calcination (clinker burning). This emission is annually reported country based to UNFCCC, using one of the three methods (Tier 1, 2 or 3). Independent of which of the methods is used, and since



the uptake takes place in the hydrated clinker components, the annual CO_2 uptake can be estimated by multiplying the calcination emission by a factor < 1. This simple methodology (named Tier 1 according to the IPCC terminology) should, as well as the emission reporting, be possible to use for all cement production. For possible uptake in other constituents than clinker, see Chapter 5.3.8 on "Cement with additions...." below. The magnitude of this factor is crucial to establish. Below, in the chapter on the use stage, a suggestion is presented.

The main constituent of cement is normally clinker, which is formed in the cement kiln. The CO₂ emissions used in these calculations are only CO₂ that is driven off from the raw material (e.g. limestone) in the cement kiln. This is referred to as calcination in this study. The CO₂ emission from the combustion in the cement kiln and other processes is not involved in the uptake calculations.

The theoretical maximum CO_2 uptake value is mainly the same as the CO_2 emissions from calcination of the raw material. This corresponds to a factor of 1.0. Such high uptake, also referred to as 100 % degree of carbonation, can eventually be found in very well hydrated concrete pieces with good access to CO_2 and moisture and after a considerable time. In reality, a practical maximum value of 75 % is reported in the European standard EN 16757 and in other literature references, for instance [11]. This value is valid for concrete, after use phase and followed by demolishing, crushing, and secondary use. It is accordingly a value that, for the bulk of concrete, takes a long time to achieve, so in spite of the long history of use, there are probably still only small amounts of concrete that has reached this status.

The practical maximum uptake value, 75 % of the emission, can only be achieved if the production of cement had been on the present level for a long time (meaning that the addition of concrete structures each year has been approximately the same and thus formed a concrete product stock that can take up CO₂ each year) and all of the structures or their demolished and crushed residue would carbonate. With an increasing amount of cement and concrete production, it is inevitable that the uptake, taking place in earlier production, will be lower than the emission from the present production.

5.2.3 The use stage of primary product

To be able to have knowledge of the annual uptake in the existing concrete products and crushed materials, one need to perform studies and calculations based on the use of concrete. This has been done in several countries and is reported in the reference list [2,3,20,5,4,13,12,6,7,8]. An overview can be found in Annex 2, Table 5. These studies can also be the basis for the development of an Advanced methodology. The methods used in different countries are all based on the well-accepted and documented carbonation rate model of square root of time dependency [11,19]. The inventories of existing concrete structures comprise different applications, exposure, and concrete quality. The age distribution of concrete in place is estimated by cement consumption statistics over time. This makes it possible to calculate a good estimate of the annual CO₂ uptake in the existing structures.

Seven such studies are compiled in Annex 2, Table 5, "Characterization of existing models..." representing Ireland, the Netherlands, Norway, Spain, Sweden, Switzerland, and "Global". The annual CO₂ uptake, for these models, in the existing concrete structures has been found to be 16, 30, 24, 3, 27, 16, and 44 % of the corresponding calcination emissions the same year.

In the referred table and references, some values can include uptake also in end-of-life and secondary use, but these values are very small, compare chapter 5.2.4, so the numbers are valid for,



and here used, for the primary products only. Some numbers are also given as percentage of total production emissions instead of percentage of calcination emissions. Those numbers have been multiplied by 1.6 to get the uptake as percentage of calcination emissions given above.

The presented numbers on uptake in relation to calcination emissions can be used to estimate a general uptake number with the help of a statistical evaluation. Before that, some of the numbers need however explanations, comments or adjustments.

The comparably low value from the **Irish study** [5] (16 %) is said to be conservatively estimated.

The value from the **Netherlands** [20] is given in the table as 19 % of the production emissions, which multiplied with 1.6 gives 30 %. It is in the report assumed that the maximum CO_2 uptake of the slag is 0.14 kg CO_2 /kg slag, corresponding to 0.14×1 440 000 = 201 600 tonnes. The maximum CO_2 uptake in slag will however, as in the clinker, not be reached. If we assume that the real uptake in the slag part is 25 kg CO_2 /tonne of slag (see Chapter 5.3.8 Cement with additions...) we get an uptake of 0.025×1 440 000 = 36 000 tonnes. This leaves 365 000 – 36 000 = 329 000 tonnes to be taken up in the clinker, corresponding to 329 000/1 318 000 = 25 %. Therefore, the clinker uptake/clinker emission is reduced from 30 to 25 %.

No adjustments are done related to the **Norwegian report** [3].

The **Spanish value** [6,7,8] of uptake related to the calcination emission is very low, only 3 %. Some reasons can be identified for the low value compared to the other studies. It is declared that the study is deliberately held on a very conservative level. The report is providing measured uptake values for laboratory samples and drilled out field samples. Uptake is estimated for a service life of 50 years for buildings and 100 years for infrastructures. The reported degree of carbonation is generally small, which for the laboratory specimens might be explained by the short time and dry testing conditions. Moreover, is it anticipated that the uptake in all interior surfaces and surfaces in contact with other materials, as well as buried concrete, is zero. The specific surface area (m²/m³) for the structures is also low, about 2 m²/m³ concrete. The large deviation from the other cement application values makes it difficult to include it in the statistical calculations below.

In a recent paper by C. Andrade: "Experimental Evaluation of the Degree of Carbonation in three Environments" [9], the earlier low values of degree of carbonation have been revised and updated, resulting in an uptake value increase to 10.8 – 11.2 % of the calcination emissions. However the assumption of zero uptake in interior surfaces and buried surfaces remain, as well as a comparably low specific surface area for the interior concrete. The motive for zero uptake in interior surfaces is that these surfaces very often are coated by for instance paint or wallpaper.

From experience and established in literature, for instance [2,3,11,13] it has been shown that CO₂ uptake takes place, even if with reduced speed, also in surfaces with some kind of coating and also under ground. See also European standard CEN/EN 16757, Annex BB.

For the purpose of this report a further recalculation of the Spanish value has accordingly been done, bringing it in line with the common view of uptake in coated surfaces and surfaces underground ("normalization").

From the Spanish report [7], table 6 it follows that 60 % of the concrete for building structures has been classified as "not exposed". 21% of it is located in foundations.



We now assume that the remaining 39 % "not exposed" are parts with some kind of cover, and, based on EN 16757, table BB1, that the uptake in such surfaces is reduced with the factor 0.7 (with cover/without cover). We also assume that the 21 % are under ground and that the uptake there is reduced with the factor 0.25 (in ground/ $\frac{1}{2}$ (exposed+sheltered)).

The buildings 60 % "not exposed", can now be calculated to correspond to $0.7 \times 39 + 0.25 \times 21 = 32.6$ % of exposed.

Further the specific surface for these, so far not included, parts is anticipated to be at least 6 m²/m³ compared to 3 for the included building surfaces. (The Fpr CEN/TR 17310:2018 presents 8 m²/m³ as a typical value for buildings, including interior parts.) The value is then increased to $(6/3) \times 32.6 = 65$ %.

This gives us for buildings $1.65 \times 10.8 = 17.8$ %. (10.8 % being the value according to C. Andrade's paper.)

From [7], table 5 it follows that 44 % of the concrete for civil works has been classified as "not exposed". It is assumed that this concrete is under ground and, as above, that it corresponds to $0.25 \times 44 = 11$ % of the above ground value.

This gives for civil works $1.11 \times 11.2 = 12.4 \%$. (11.2 being the value according to C. Andrade's paper.)

The relation in Spain between concrete used for buildings and for civil works is about 47/53. (Cement consumption statistics from Cembureau 2016.)

We now get for all concrete structures $0.47 \times 17.8 + 0.53 \times 12.4 = 14.9 \%$

The Spanish CO₂ uptake value used for the statistical calculation below; is accordingly 15 % of the calcination emissions.

No adjustments are done related to the **Swedish report** [2].

In the **Swiss report** [13], the degree of carbonation is set to 75 % throughout, but is later in another paper [12] changed to 50 %. An average uptake of 10 % of the total emissions, for 50 years is presented in the report. For the calcination emissions, the figure can be estimated to be $10\times(1/0.63)=16$ %. For 100 years, the figure can be estimated to be $16\times\sqrt{(100/50)}=23$ %. With degree of carbonation being 50 % instead of 75 %, we get $23\times(50/75)=15$ %. (According to the Annex BB of the EN 16757, the degree of carbonation varies between 40 % for indoor structures to 85 % for outdoor structures not sheltered from rain.)

The **global value** [4] of 44 % should be adjusted due to the very large amount of mortar included in the study. The mortar contribution to the CO_2 uptake is about 70 % of the global study value 44 %, though the amount of cement used for this application is about 30 %. The concrete and mortar uptake figures can be calculated as follows.



Assume that the uptake factor for concrete is c and the factor for mortar is m.

The factor for concrete and mortar together is 0.44. Then

 $70c + 30m = 100 \times 0.44$

 $c/m = 30 \times 30/70 \times 70 = 0.18$ (70 % of concrete accounts for 30 % of the uptake and vice versa for mortar.)

 70×0.18 m + 30m = 44

m = 44/42.6 = 1.03

 $c = 0.18 \times 1.03 = 0.19$

The uptake factor for concrete is thus 0.19, (19 %).

(It should be noted that this calculation, with reduction of the mortar portion, shows that the high global value of 44 % is reduced to a more typical value of 19 %, which strengthen the credibility of the global study.)

After these comments, the series of numbers of CO₂ uptake related to calcination emissions to be used for the statistical evaluation are: 16, 25, 24, 27, 15, 15 and 19 %.

If we perform a very simple statistical analysis of the results, assuming normal distribution, we end up with a mean value of 20 % and a standard deviation of 5.1 %. The value 15 % accordingly corresponds to (Mean - $1 \times St$. dev.) and is a value with some safety margin.

It should be noted that the statistical basis is only six country values. The base for the six single numbers varies also for instance with statistical accuracy and with different cement types. The studies covers however a large number of structures and products.

With a mean cement consumption as basis, instead of a single year value, the accuracy of the relative uptake values presented for the different countries should be more accurate. Below, the mean cement consumption during the years 2011-2016, Table 1, is used as basis for the calculation of the relative CO_2 uptake value for five countries.

Table 1	Yearly cement consumption in kton 1.
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Country	2011	2012	2013	2014	2015	2016	Average (μ)	Standard deviation (σ)	Coefficient of variation (σ/μ)
Ireland	1 310	1 130	1 064	1 195	1 720	1 820	1 373	319	0.23
Netherlands	5 596	5 079	3 991	4 100	4 040	4 110	4 486	681	0.15
Norway	1 874	1 943	1 993	1 967	1 960	2 060	1 966	61	0.03
Sweden	2 440	2 340	2 235	2 200	2 270	2 360	2 308	89	0.04
Switzerland	5 299	5 014	5 274	5 370	4 900	5 000	5 143	195	0.04

The so far presented numbers (based on the single year values) for the five first countries are 16, 25, 24, 27 and 15%.

 $^{^{1}}$ Data from VDZ (Verein Deutscher Zementwerke e.V.). VDZ is the economic, technical and scientific association for the German cement industry.



The single year values used for the calculation of these numbers are the following:

Ireland: 2000 ktonne Netherlands 4040 ktonne Norway: 1550 ktonne Sweden: 2440 ktonne Switzerland: 4553 ktonne

(The actual year lies sometimes before 2011 or has been modified in the referred country report. The single year value used for Spain is not known so no calculation has been made.)

If we chose to use the six years mean value instead we get the following corrected numbers:

Ireland $(2000/1373) \times 16 = 23 \%$ Netherlands $(4040/4486) \times 25 = 23 \%$ Norway $(1550/1966) \times 24 = 19 \%$ Sweden $(2440/2306) \times 27 = 28 \%$ Switzerland $(4553/5142) \times 15 = 13 \%$

With these new numbers, we get, including the Spanish value 15 and the global value 19 % unchanged:

Mean value: 20.0 % St. dev.: 5.13 %

These values are very similar to the ones we got with the single year basis.

The mean value is, as mentioned above, 20 %, (Tier 1A).

In order to estimate the annual uptake with some safety margin, it is accordingly suggested that the annual emissions could be multiplied with the factor 0.15 when handling the uncertainties in the Simplified methodology (Tier 1B).

It is likely to believe that the Simplified methodology will be used in most countries, at least in the beginning of the calculation of CO₂ uptake in concrete and other cement containing products. However, due to the simplifications of the calculations used in the method, the uncertainty will be relatively high and the accuracy of the national calculations can be poor. The uptake factor can, in reality, vary between different countries, which a general uptake factor cannot account for. Technical factors such the use of mortar can have a significant impact. However, on a global scale, the average uptake value will be reflected by the general uptake factor.

5.2.3.1 Influence of high use of mortar (from the global study)

The following indicates a method to consider a high mortar use in the cement applications. It is based on the global study [4] of the "Characterization table in annex 2, Table 5".

The mortar contribution to the CO₂ uptake is about 70 % of the global study value of 44 % carbonation, though the amount of cement used for this application is about 30 %.

The concrete and mortar uptake figures can be calculated as follows:



Assume that the uptake factor for concrete is c and the uptake factor for mortar is m. The factor for concrete and mortar together is 0.44. Then

```
70c + 30m = 100 \times 0.44

c/m = 30 \times 30 / 70 \times 70 = 0.18 (70 % of concrete accounts for 30 % of the uptake and vice versa for mortar.)

70 \times 0.18m + 30m = 44

m = 44 / 42.6 = 1.03

c = 0.18 \times 1.03 = 0.19
```

The uptake factor for concrete is thus 0.19, which seems to be a reasonable value.

The uptake factor for mortar is very high, 1.0. It means that all cement used for mortar applications is totally carbonated already after one year. This result is also apparent from the global report [4] as a result of thin structures (\approx 20 mm), high rate of carbonation (k = 19.6) and high degree of carbonation (average 92 %). This can be explained by a high use of mortar for rendering.

The factors 0.15 or 0.20 are suggested above to be valid for more real concrete applications, say with a mortar use of less than 10 %. If the mortar use is larger than 10 %, an adjustment factor could possibly be used. With the mortar share being 30 %, the uptake factor is 0.44. If one then assume that each 10 % increase of mortar, increases the uptake factor with 0.10 we obtain:

Alternative A (Tier 1A)

With 0-10% of cement use in mortar for rendering applications, the uptake factor will be 0.20.

With 10 - 30 % of cement use in mortar for rendering applications, the uptake factor will be: 0.20 + 0.0115 (MR - 10), where MR is the Mortar percentage for rendering.

Alternative B (Tier 1B)

With 0-10% of cement use in mortar for rendering applications, the uptake factor will be 0.15.

With 10 - 30 % of cement use in mortar for rendering applications, the uptake factor will be: 0.15 + 0.01 (MR - 10), where MR is the Mortar percentage for rendering.

5.2.4 End-of-life stage and secondary use

In addition to the CO₂ uptake over the use stage of structures, there is uptake in the end-of-life stage and the secondary use, normally as crushed material. Detailed information on the amount of this material and its uptake is rare, so it is today difficult to give a reliable estimate. The percent of concrete recycling, given in some countries is unfortunately not sufficient, since it is mainly the service life at demolition, the processing of the demolished material and the applications for the crushed concrete that determines the actual uptake.

5.2.4.1 End-of-life stage (demolishing, crushing and storage)

The EN 16757, Annex BB [16] indicates a conservative value of a CO_2 uptake in the end-of-life stage (demolishing, crushing, and storage) of $5 \text{ kg } CO_2/\text{m}^3$ of concrete with present waste handling systems. With an average cement content of 330 kg/m³ concrete, this corresponds to approximately $5/0.330 = 15 \text{ kg } CO_2/\text{tonne}$ of cement or 15/490 = 3% of a Portland cement calcination emission. The figure 5 kg CO_2/m^3 of concrete indicates a very small uptake in this stage. The potential to increase this value by applying a more advanced waste processing systems is however very large.



The **Swedish study** [2] reports that about 1.5 million tonne (625 000 m³) of concrete is annually demolished. The annual uptake in the end-of-life stage and secondary use is presently estimated to be only 12 000 tonne and of that, about 4000 tonne of the 12 000 tonne of uptake after the use stage takes place in the end-of-life stage. This corresponds to $4\,000\,000/625\,000 = 6.4\,\mathrm{kg}\,\mathrm{CO}_2/\mathrm{m}^3$ of concrete or $6.4/0.33 = 19\,\mathrm{kg}\,\mathrm{CO}_2/\mathrm{tonne}$ cement, or $19/490 = 4\,\%$ of a Portland cement calcination emission.

Since the annual amount of concrete taken out of the service life is rather small compared to the annual production, $625\,000/7\,000\,000\,\text{m}^3 = 9\,\%$, the uptake is small in <u>relation to the annual calcination emissions</u>, only about $0.4\,\%$, $(0.09 \times 0.04 \times 100)\,\%$.

The **"global"** study [4] value 44 % includes uptake in the end-of-life stage. The average exposure time is 0.4 years. The relative uptake in this stage is reported to be around 8 % (1.4/(16.1+1.4)) in concrete and around 2 % in mortar.

With about 70 % concrete and 30 % mortar we end up with a mean figure of about

 $0.70 \times 0.08 + 0.30 \times 0.02 = 0.06$

Thus, about 6 % of the uptake takes place in the end-of-life stage. This corresponds to a factor $0.06\times0.44 = 0.026$ or 2.6 % of the annual calcination emissions. The total calcination emission being 0.55 Gt C, corresponding to $(44/12)\times0.55 = 2.02$ Gt CO₂. The 2.6 % means an uptake in the end-of-life stage of $0.026\times2.02 = 0.0525$ Gt or 52.5 million tonne of CO₂. The annual amount of concrete entering the end-of-life stage is not found in the report. The uptake can therefore not be presented as kg CO₂/m³ of concrete.

The **Norwegian study** [3] reports that the "concrete demolition rate is 10 % of the consumed concrete volume" and that 90 % of the demolished concrete "recovery rate" is available for carbonation. No figures are specified for the end-of-life stage. For a 100 years "recovery phase", a scenario is however presented, see Secondary use.

In **the Netherlands** [20], the amount of annual demolished concrete as a percentage of annual production is high, 5.1 million m³ out of 14 million m³, corresponding to 36 %. The Dutch study reports an uptake in the end-of-life stage of 83 000 tonne per year. The calcination emission is 1 960 0000 ton, which means that the uptake in the end-of-life stage is reported to be 83 000/1 960 000 = $\frac{4}{6}$ % of the calcination emissions. Slag cement is here a considerable part of the cement consumption. The uptake per m³ of demolished concrete is 83 000 000/5 100 000 = $\frac{16}{6}$ kg $\frac{16}{6}$ kg $\frac{16}{6}$ concrete.

Measurements of CO_2 uptake in crushed concrete from demolition sites and recycling plants are reported in (Kikuchi T and Kuroda Y, 2011) [10]. The presented figure of the CO_2 uptake in a couple of months is 11 kg per tonne of crushed concrete, corresponding to about $25 \text{ kg } CO_2/\text{m}^3$ of concrete, or to about $25/0.330 = 76 \text{ kg } CO_2/\text{tonne}$ cement, or to 76/490 = 15 % of the calcination emissions for the cement in the crushed concrete, (Portland cement anticipated). The relation of the uptake at the end-of-use stage, to the annual total calcination emission, can be calculated if the annual recycled amount and the production can be found. The annual amount of CO_2 uptake by "recycled crusher run" is estimated to 0.37 million tonne. The annual produced amount of concrete is 500 million tonne, or about 200 million m^3 . The average cement content is assumed to be 330 kg/m³, and the calcination emission 0.480 kg CO_2 per kg cement. The emission per m^3 concrete will be $330 \times 0.480 = 158 \text{ kg } CO_2/m^3$ concrete. The relative uptake is accordingly:

 $370\ 000/200\ 000\ 000 \times 0.158 = 0.012$ or $1.2\ \%$ of the total calcination emissions.



Considering that the concrete that is taken out of service annually is normally only a minor part of the annual production, it means that the uptake in this stage compared to the emissions is normally small. This means also that any default factor value of the calcination emissions in the Simplified methodology need to be small, (e.g. range of 0.4 - 4 %). If there however exists knowledge of the annual amount of concrete, taken out of service, (often referred to as recycled concrete) the uptake in this concrete during the end-of-use stage can be estimated to be in the range of $5 - 25 \text{ kg CO}_2/\text{m}^3$ of concrete.

For the Simplified methodology, it therefore seems reasonable to suggest two principal ways to report an estimated uptake value in the end-of-life stage:

If the annual amount of concrete in the end-of-life stage is known (m³ per year), the uptake can be calculated as 10 kg CO₂/m³ of concrete.

If the annual amount of concrete in the end-of-life stage is not known, the uptake can be calculated as 2 % of the annual calcination emissions, (factor 0.02). This also includes an anticipated minor increase of concrete to end-of-life the coming years.

Potential uptake in the end-of-life stage (demolition, crushing, and stockpiling) is very large if adequate measures are taken. The measures are storage of the crushed material in fractions and sheltered from rain in order to enhance air circulation in the piles, [17] and draft FprCEN/TR 17310, Carbonation and CO₂ uptake in concrete. See also the Advanced methodology.

5.2.4.2 Secondary use

The uptake in secondary use, mainly in various applications for crushed concrete, is probably of great importance but very few reports present reliable uptake in this case. The secondary use phase reminds a lot of the primary use phase. By crushing or other processing, new concrete surfaces for carbonation are exposed while old surfaces and volumes can continue to carbonate. Thus, similar methods as for primary concrete products can be utilized.

The **Swedish study** [2] estimates that about 8000 of the 12 000 tonne of the uptake after the use stage takes place during the "use of the mixed crushed concrete fractions", that is as secondary use. This corresponds to 8 000 000/625 000 = $12.8 \text{ kg CO}_2/\text{m}^3$ of concrete, or $12.8/0.33 = 39 \text{ kg CO}_2/\text{tonne}$ cement, or (39/490)*100 = 8 % of a Portland cement calcination emission. Since 1.5 million tonne (625 000 m³) of concrete is recycled each year and the production is about 7 000 000 m³, the relative recycling rate is 9 %. This means that the annual uptake in the recycled concrete is less than 1 % of the annual calcination emissions, $(0.09 \times 0.08 \times 100 = 0.72 \%)$.

The **Norwegian study** [3] assumes a $\underline{100 \text{ years}}$ "recovery stage". The scenario is an anticipated crushed concrete particle size distribution and 10 % exposed above ground and 90 % below ground. The annual uptake is estimated to be 25 000 tonne, corresponding to 25 000/795 000 = 3 % of the annual calcination emissions. Counted as uptake related to the recovered material (1 million tonne or 400 000 m³) the figure is 25 000 000/400 000 = $\underline{62 \text{ kg CO}_2/\text{m}^3 \text{ concrete}}$. This figure is actually the annual uptake in "recovered concrete" during 100 years, or the annual uptake 2011, assuming 100 years of 10 % recovery rate each year.

The **Dutch study** [20] doesn't present any values for the secondary use.

The **global study** [4] claims that only 0.1 % of the initial emission is absorbed during the disposal or reuse of the concrete waste. This corresponds to about $0.001\times0.49\times330 = 0.16$ kg CO_2/m^3 of



<u>concrete</u>. It is uncertain however if "reuse of concrete waste" is the same as secondary use, so this very low value is, until more information is available, not taken into account.

To summarize: As in the "end-of-use stage", only a minor amount of the annually produced concrete is <u>presently</u> normally taken out of service and being recycled. The uptake in the secondary use applications is for this reason small compared to the emissions of the produced concrete and cement.

The default value in the Simplified methodology (without any other data than calcination emissions) must therefore be modest. Proposed is 0.01 or 1 % of the calcination emissions.

For the Simplified methodology, it therefore seems reasonable to suggest two principal ways to report an estimated uptake value in the secondary use:

If the annual amount of concrete entering the secondary use is known (m³ per year), the uptake can be calculated as 10 kg CO_2/m^3 of concrete. This value is valid for unbound applications of crushed material.

If the annual amount of concrete entering the secondary use is not known, the uptake can be calculated as 1 % of the annual calcination emissions, (factor 0.01).

5.2.5 Summary

The annual CO_2 uptake of concrete in an area or country – in use stage (existing structures), end-ofuse stage (demolition, crushing, stockpiling), and secondary use – can be estimated according to the Simplified methodology. The uptake values are related to the reported calcination emissions of the <u>consumed clinker</u> (produced-export+import) in the same area or country. Note that the nationally calculated emission values may need to be adjusted for exports and imports of cement/clinker.

Use stage

Two alternative values, A and B, for the CO₂ uptake in the use stage are given. The reason for giving two values is to give options how to handle the uncertainty of the calculated figures. From Chapter 5.2.3 it follows that one can present a calculated mean value (A) or a low (Mean - St. dev.) value (B). The IPCC recommendation is to not overestimate, nor underestimate in the calculations, which should indicate alternative A. Uncertainties can also be handled by moving to higher Tierlevels.

Alternative A:

The annual uptake in the use stage can be estimated as 0.20×(the reported emission from calcination of consumed cement clinker).

If the mortar for rendering applications, in total, amount to more than 10 % but less than 30 % of the cement consumption, the annual uptake factor in the use stage can be estimated at 0.20 + 0.0115(MR - 10), where MR is the mortar percentage for rendering.



Alternative B:

The annual uptake in the use stage can be estimated as 0.15×(the reported emission from calcination of consumed cement clinker).

If the mortar for rendering applications, in total, amount to more than 10 % but less than 30 % of the cement consumption, the annual uptake factor in the use stage can be estimated at 0.15 + 0.01(MR - 10), where MR is the mortar percentage for rendering.

End-of-Life stage and secondary use

Annual uptake in the end-of-life stage and secondary use can be estimated at $(0.02 + 0.01) \times ($ the reported emission from calcination of consumed cement clinker)

Alternatively, the following estimation can be done in the end-of-life stage and the secondary use.

- If the annual amount of concrete being taken out of service and processed on a recycling plant is known, the CO₂ uptake in the end-of-life stage can be calculated to 10 kg CO₂/m³ concrete.
- If the annual amount of crushed concrete, entering the <u>secondary use</u> as unbound material, is known, the uptake can be calculated to 10 kg CO₂/m³ concrete.

Examples:

For 2 million tonne of Portland cement:

The CO_2 emission is 2 000 000×0.490 = 980 000 tonne.

Alternative A:

The CO₂ uptake in use stage, end-of-life stage and secondary use is $(0.20 + 0.02 + 0.01) \times 980~000 = 225~400$ tonne.

Alternative B:

The CO_2 uptake in use stage, end-of-life stage and secondary use is $(0.15 + 0.02 + 0.01) \times 980~000 = 176~400$ tonne.

5.3 Tier 2 CO₂ uptake model - Advanced methodology

5.3.1 Proposed CO₂ uptake model for Tier 2

The present model, representing Tier 2, provides a more advanced calculation method for estimating the annual uptake of CO₂ in existing concrete structures on a national basis. The model should be used primarily in cases where resources are missing to perform a more accurate calculation according to the calculation methods described for Tier 3 of Chapter 5.4, yet it has the potential and knowledge to perform a more advanced and improved CO₂ uptake model than the simplified model offered in Tier 1. According to IPCC rules, Tier 2 and 3 should also replace Tier 1



if the uptake in the country is defined as a "key category". However, in this model, a more simplified calculation of the CO₂ uptake in existing (old) concrete structures is used compared to Tier 3 and referred to as Onward calculation method. The proposed methodology for Tier 2 is based on methodology II described below. The national annual CO₂ uptake in concrete; in use stage (existing structures), in end-of-use stage (demolition, crushing, stockpiling), and in secondary use, can be estimated according to this methodology for Tier 2.

5.3.1.1 Primary use stage

The mean value of at least 10 years of cement consumption (all available data shall normally be used) may be used together with knowledge of one year of concrete use. The mean value of the cement consumption should normally be easy to access and would give a better estimation of the amount of carbonating structures than only the last year's consumption. It is anticipated that the uptake during one year in existing buildings is equal to the uptake during (100) years in the concrete produced during the same year. For more information about this "onward" calculation method, see Chapter 5.3.3.2.1.

A suggested general description of the required steps in the proposed Advanced methodology II is presented below, based on parts of the Swedish study [2] above and the paper by C. Andrade [9].

➤ Identify typical concrete applications (frequent type of structures). Normally, at least 5 applications should be needed, corresponding to at least 65 % of the cement consumption. For instance: Bridges, residential buildings, office buildings, roof tiles, pavement, shotcrete, sleepers, and mortar. Applications outside the chosen ones are treated as the most similar of the defined ones.

Cement content, concrete quality, exposure and specific surface (m² surface/m³ concrete) is described for each application. Especially exposure and specific surface may vary within the application.

Calculate the CO₂ uptake per m³ concrete for each application as a function of time:

The CO₂ uptake in kg per m³ concrete for each application during t years can be calculated as: (The formula can be found in EN 16757, Annex BB.)

$$CO_2$$
-uptake = $(\Sigma(k_i \times DOC_i \times A_i)) (\sqrt{(t)/1000}) \times U_{tcc} \times C$

Where

 CO_2 -uptake is the total CO_2 uptake in kg CO_2/m^3 concrete.

 \mathbf{k}_{i} is factor for the rate of carbonation for surface i in mm/ $\sqrt{(t)}$.

DOC_i is the degree of carbonation for surface i.

 A_i is the area of surface i in m^2 .

t is the number of years.

 U_{tcc} is the maximum theoretical uptake in kg CO₂/kg cement. The value is ≈ 0.49 for Portland cement (CEM I).

C is cement content in kg cement/m³ of concrete.

Values of k can be found in EN 16757, Annex BB.

Values of DOC can, for example, be found in EN 16757, Annex BB.



Table 2 Table with k-factors for calculation of depth of carbonation for different strength classes (cylinder) and exposure conditions and also degree of carbonation for different exposure conditions. This table is reproduced from Annex BB of European standard EN 16757:2017, [16].

Concrete strength	≤15 MPa	15-20 MPa	25-35 MPa	≥ 35 MPa	Degree of carbonation (DOC)
Parameters	Va	Percentage			
Civil engineering structures					
Exposed to rain		2.7	1.6	1.1	85
Sheltered from rain		6.6	4.4	2.7	75
In ground *		1.1	0.8	0.5	85
Buildings					
<u>Outdoor</u>					
Exposed to rain	5.5	2.7	1.6	1.1	85
Sheltered from rain	11	6.6	4.4	2.7	75
Indoor in dry climate ***					
With cover **	11.6	6.9	4.6	2.7	40
Without cover	16.5	9.9	6.6	3.8	40
<u>In ground *</u>		1.1	0.8	0.5	85

^{*} Under groundwater level k = 0.2

Examples of this kind of calculation can be found in EN 16757, Annex BB, example 5 and 6.

- Calculate the last 20 years mean annual cement consumption and estimate the annual concrete production and the distribution on the different applications.
- Calculate the CO₂ uptake for each application during 100 years. (normally)
- ➤ Calculate the sum of the annual CO₂ uptake of all concrete applications.
- > The sum of the cement content in the produced concrete should always be checked against the cement production.

5.3.1.2 End-of-life stage (demolishing, crushing and storage)

In the Simplified methodology, we introduced a conservative "default" value for this phase based on the present normal handling procedures of demolished and crushed concrete. These procedures include normally storage in large unsheltered piles, during a rather short period of time. Moreover, the recycling rate (the annual amount of demolished and crushed concrete in relation to the annual production) is normally low in most countries. The volumes can however be anticipated to increase in the future, since more concrete structures reach the end of the service life. It is therefore important to base the uptake calculations on the real amounts of concrete to end-of-life handling.

The following estimations for CO₂ uptake in the end-of-life stage can be given.

For normal handling procedure or recycling rate less than 5 %: (Same as for Simplified methodology, Tier 1)

^{**} Paint or wall paper. (Under tiles, parquet and laminate k is considered to be 0.)

^{***} Indoor in dry climate means that the RH is normally between 45 and 65 %.



If the annual amount of concrete in the end-of-life stage is known (m³ per year), the uptake can be calculated as 10 kg CO₂/m³ of concrete.

If the annual amount of concrete in the end-of-life stage is not known, the uptake can be calculated as 2 % of the annual calcination emissions, (factor 0.02).

For improved handling procedure:

A preliminary suggestion is that the uptake could be set to 20 kg CO₂/m³ of concrete if an enhanced procedure with air access in the fractions and at least 4 months storage in at least three fractions is applied. In this case, the amount of concrete need to be known.

5.3.1.3 Secondary use

Only very small amounts of concrete structures that have reached their service life, actually ends up as waste (landfill), but are used in crushed form as, or in, a new product – secondary use. It is therefore important to base the uptake calculations on the real amounts of concrete to secondary use. The CO₂ uptake in secondary use is quite similar to the uptake in primary use, so similar calculation methods could be used. However, even more factors are unknown for secondary use, so it can be difficult to create general but accurate methods. The more exact methods are often quite specific and depend on the type of secondary use (country specific), and may be treated under Tier 3 in Chapter 5.4. Here, a more general method is proposed.

Under favourable conditions for the secondary use applications, the <u>total uptake (primary use + end-of-life + secondary use)</u> can amount to about 75 % ([11], [16]) of the maximum theoretical potential (equal to the calcination emission), corresponding to about 110 kg CO_2/m^3 for an average concrete.

As a lowest level for secondary use, an uptake according to the Simplified method (Tier 1) can be used.

That is:

If the annual amount of concrete entering the secondary use is known (m^3 per year), the uptake can be calculated as 10 kg CO_2/m^3 of concrete. This value is valid for unbound applications of crushed material. (If the annual amount of concrete entering the secondary use is not known, the uptake can be calculated as 1 % of the annual calcination emissions, factor 0.01.)

5.3.1.4 Cement constituents and concrete additions

Different cement constituents and concrete additions are frequently used in different parts of the world, but the degree of use varies considerably between different manufacturers and countries. The materials used are also very different and their ability to bind CO₂ is also very different. The additions that are mainly used are ground granulated blast-furnace slag (GGBS), fly ash, silica fume, and limestone. Relatively few data are available for the uptake of CO₂ in different additions, but some data, especially for blast-furnace slag (GGBS), exist. The proposal is therefore to include only CO₂ uptake in slag (GGBS) for Tier 2 and a more detailed model for CO₂ uptake in various additions can then be made in Tier 3. In Tier 3, specific CO₂ uptake values for different additions can be obtained and used in the models.



The proposed CO₂ uptake value for ground granulated blast-furnace slag (GGBS) is 25 kg CO₂/tonne GGBS, see further Chapter 5.3.8.

The following chapters explain the background and consideration that underlie the proposed model.

5.3.2 Introduction

The Advanced methodology (AM) should be used to make a better estimate of the annual CO₂ uptake than with the Simplified methodology (SM). As mentioned before, it normally gives a more accurate uptake than the simplified one, thus encouraging the extra effort.

To be able to make a good estimate of the annual CO₂ uptake, one need a good overview and knowledge of the existing concrete product stock in the country or region in question, since it is in these structures that the uptake takes place.

There is of course a correlation between the cement consumption and the concrete production. The best way to obtain a historical estimate of the concrete production is normally via the cement consumption, whose statistics are often available. For the Advanced methodology, it is essential that the sum of the cement content in the produced concrete is checked against the cement consumption. The cement consumption includes cement production as well as import and export of cement.

The basis for the Advanced methodology (as well as for the Simplified methodology) is accordingly the cement consumption. Here, we should however have knowledge of, not only the consumption of the year in question, but a historical view going at least 10 years back, but preferably more, in order to have a reasonable picture of the volume of the concrete production and thus the concrete product stock.

In lack of data, it is in IPCC accepted to use extrapolation, expert opinion and other quantification methods.

5.3.3 The primary use stage, results from different studies

The distribution of the cement consumption on different applications, for instance infrastructure, residential buildings, and other buildings, with typical surface/volume ratio and concrete quality, should be known. It has been shown [2] that it might be sufficient to know this distribution, not every year, but on a couple of occasions. Since the CO₂ uptake takes place in concrete produced during a long time, the relative changes in cement distribution for different applications, due for instance to economic reasons, seem to even out in the long run. It is thus likely more important to have a detailed knowledge of the cement distribution on different applications and products and of their exposure and surface/volume ratio on one occasion, than a vague picture at many occasions.



The different methods listed in the table "Characterization of existing models...in Appendix 1, Table 5" have somewhat different approaches to the estimation of the annual CO₂ uptake. They are below arranged under two headlines, Advanced methodology I and II.

Two of the methods estimate the annual uptake in the existing building stock by using extensive knowledge of the historical cement use, (Advanced methodology I). These two are also the base for the later Tier 3 method.

Four of the methods estimate the annual uptake in the concrete product stock by using the assumption that the uptake during one year in concrete that was produced during the previous (100) years, which is the desired information, can be placed on an equality with the uptake the coming (100) years in concrete produced during the same year, (Advanced methodology II). This is referred to as the "Onward calculation method". The uptake in the existing concrete product stock during a specific year (the reporting year) is what is requested. This uptake can be calculated by using knowledge of the historical concrete production (100 years if possible) as in Advanced methodology I. This knowledge is, however, often not at hand and the uptake calculation is also laborious. One can instead make use of the assumption that this "true" uptake value can be estimated by the uptake the coming (100) years in concrete produced the same year (the reporting year). How accurate this estimation is depends a lot on how stable the use of concrete has been over time. See further Chapter 5.3.3.2.1.

The table below is a short survey of the six different methods.

Table 3 The table shows the different calculation principles for the different country specific CO₂ uptake models compared to Tier 1A (Mean) and Tier 1B (Mean – St. dev.).

Country/method	Advanced	Advanced	Simplified	Simplified	
	methodology I	methodology II	methodology Tier 1A	methodology Tier 1B	
			Relative values	Relative values	
Sweden	Yes		84 %	65.5 %	
Norway		Yes	112 %	88 %	
The Netherlands		Yes	86 %	64 %	
Ireland		Yes	125 %	94 %	
Global	Yes		106 %	87 %	
Switzerland		Yes	133 %	100 %	
Spain		Yes	133 %	100 %	

The seven methods in both Advanced methodology I and II have a similar approach when it comes to knowledge of cement distribution on different concrete applications/products. That is their relative amount, exposure, mean surface/volume ratio, concrete quality, and cement content. The CO₂ uptake of each application is calculated using k-factors and the same kind of square root of time dependency. Value of degree of carbonation may differ.



5.3.3.1 The Advanced methodology I

<u>The Swedish method</u> [2] uses knowledge of 100 years cement consumption and 60 years of distribution on different applications and knowledge of how concrete structures are distributed throughout the building sector to estimate the stock of concrete applications. The k-values and the degree of carbonation are very similar to the ones in EN 16757 [16]. The uptake each year is calculated as the difference between two consecutive years and is summed up for all the 7 applications/products.

Influence of uncertainties in the model

The result is obtained by using the market distribution of the concrete applications valid for the year 2010. A calculation based on actual historical statistics between 1950 and 2010 shows very similar results. The results show a dependence on the concrete quality, which is expected. If hypothetically all concrete would be of 45 MPa, the uptake would be reduced from about 300 to 170 ktonne. With 25 MPa, the uptake would increase to 470 ktonne. In practise, these extreme cases will not occur. Applying different lengths of service life (70-100 years) only results in minor changes to the uptake. The model input data, which consist of cement production and cement use in various products as well as cement types and concrete qualities, has both high availability and good reliability in most countries. The model has been designed to be robust to historical variations. The quality of the results is mainly influenced by the quality of the input data and its statistical distribution for the most recent year, compare [21].

This way of calculating could be characterized as an extra advanced methodology, where concern is taken to the historical cement/concrete production during 100 years, and also the distribution of cement on different applications is known under 60 years. The calculation should give a very good estimate of the annual uptake. With this methodology, it has been estimated that the CO₂ uptake in Sweden during the year of 2011 amounted to 300 000 tonne.

Comparison with Simplified methodology Tier 1A

If the Simplified methodology would be used, the result would be:

Calcination emission 455 kg CO₂/tonne cement (A mixture of CEM I and CEM II/A)

Cement consumption: 2.4 million tonne

 CO_2 uptake is $0.23\times0.455\times2~400~000 = 251~160$ tonne, corresponding to 83.7 % of the advanced methodology value.

Comparison with Simplified methodology Tier 1B

If the Simplified methodology would be used, the result would be:

Calcination emission 455 kg CO₂/tonne cement (A mixture of CEM I and CEM II/A)

Cement consumption: 2.4 million tonne

 CO_2 uptake is $0.18 \times 0.455 \times 2400000 = 196560$ tonne, corresponding to 65.5% of the advanced methodology value.

The global method [4] The data is provided for different regions (China, US, Europe and rest of the world). Degree of carbonation is set to 80 % in concrete and 92 % in average of mortar. The method uses 83 years (1930 – 2013) of cement consumption and a variety of regional information on concrete applications. The CO_2 uptake in the relation to the calcination emissions is given as a mean value for the 83 years, being 43 %, and for the year 2013, being 44 %. To be noted is the high share of cement for mortar application; 30 % is used in this application, and the uptake share is 70 %, due to the high surface/volume ratio and relatively low quality of the mortar. The uptake during the year 2013 is reported to be 0.24 Gt C, corresponding to $(44/12)\times0.24 = 0.88$ Gt, or 880



million tonne of CO₂. This figure includes the end-of-life stage. The calcination emission during 2013 is reported to be 0.55 Gt C, corresponding to 2.02 Gt CO₂.

Comparison with Simplified methodology Tier 1A

With the suggested Simplified methodology the uptake should have been: (With 30 % in mortar applications.)

The uptake factor is 0.20 + 0.0115(30 - 10) + 0.02 + 0.01 = 0.20 + 0.23 + 0.02 + 0.01 = 0.46. CO₂ uptake is $0.46 \times 2~020 = 929$ million tonne, corresponding to 929/880 = 106~% of the advanced methodology value.

Comparison with Simplified methodology Tier 1B

With the suggested Simplified methodology the uptake should have been: (With 30 % in mortar applications.)

The uptake factor is 0.15 + 0.01(30 - 10) + 0.02 + 0.01 = 0.15 + 0.20 + 0.02 + 0.01 = 0.38. CO₂ uptake is $0.38 \times 2.020 = 768$ million tonne, corresponding to 768/880 = 87% of the advanced methodology value.

5.3.3.2 The Advanced methodology II

Other methods in the country overview make use of the assumption that the uptake during one year in concrete that was produced during the previous (100) years, which is the desired information, can be placed on an equality with the uptake the coming (100) years in concrete produced during the same year. This can be characterized as "onward" calculation, which is further explained in Chapter 5.3.3.2.1.

For a 100 % identity, it is required unchanged production of cement and concrete and distribution on applications during the previous years. In practise, this unrealistic requirement can however be modified.

To be noted is that the years immediately prior to the reported year are most important for the uptake amount. This results from the square root of time dependency of the carbonation rate, which means that the concrete production of remote years becomes less important.

The Norwegian method [3]

The degree of carbonation is set to 70 % all over. The Norwegian report doesn't make any calculation based on historical values of cement consumption. The output of the model is the total amount of CO₂ bound by concrete consumed in Norway in 2011, assuming an onward service life of 100 years. It is anticipated that the uptake during one year in existing buildings is equal to the uptake during 100 years in the concrete produced during one year. Thus, in this case, only the production of cement and concrete applications during one year need to be known.

With this method, it is estimated that the annual uptake during 2011 in Norway is 140 000 tonne in the service life and 165 000 tonne if 100 years of "recovery phase" (secondary use) is added. In total, an uptake of 305 000 tonne CO_2 .

(It is however also reported that the uptake is 94 kg, resp. 111 kg CO_2 /tonne cement. With the annual cement consumption being $\approx 1\,800$ ktonne, it means an annual uptake of 171 000, resp. 202 000 tonne. The apparent inconsistency depends on that the CO_2 uptake is calculated on basis of



the actual concrete consumption, which corresponds to a "fictive" cement consumption of ≈ 1500 ktonne.)

From the Norwegian report, it can be found that three different cements (CEM I, CEM II/A-V and CEM II/B-S) have been used. The maximum CO_2 binding 2011 (70 % of the calcination emissions) is respectively 215, 252, and 55 thousand tonne of CO_2 , in total 557 000 tonne. The calcination emissions are accordingly 557 000/0.70 = 795 000 tonne.

Comparison with Simplified methodology Tier 1A

With the suggested Simplified methodology, we should get in the use stage:

 $0.20 \times 795\,000 = 159\,000$ tonne bound in the clinker

The GGBS amount in the CEM II/B-S (33 % GGBS) is $0.33 \times 186\ 000 = 61\ 000$ tonne.

The uptake will accordingly be $61\ 000\times0.035 = 2135$ tonne (35 kg CO₂/tonne of GGBS, compare 5.3.8, last paragraph)

Total uptake in the use stage by the simplified method is 161 135 tonne compared to 140 000 tonne with the Advanced methodology (115 %).

The End-of-life stage and secondary use would add an extra $0.03 \times 795\,000 = 23\,850$ tonne to the uptake, in total 185 000 compared to 165 000 with the Advanced methodology (112 %).

Comparison with Simplified methodology Tier 1B

With the suggested Simplified methodology, we should get in the use stage:

 $0.15 \times 795\ 000 = 120\ 000$ tonne bound in the clinker

The GGBS amount in the CEM II/B-S (33 % GGBS) is $0.33 \times 186\,000 = 61\,000$ tonne.

The uptake will accordingly be $61\ 000 \times 0.025 = 1\ 500$ tonne (25 kg CO₂/tonne of GGBS)

Total uptake in the use stage by the simplified method is 121 500 tonne compared to 140 000 tonne with the Advanced methodology (86 %).

The End-of-life stage and secondary use would add an extra $0.03 \times 795\,000 = 23\,850$ tonne to the uptake, in total 145 350 compared to 165 000 with the Advanced methodology (88 %).

The Dutch method [20]

The amount of GGBS in the different applications/products is provided. It is assumed that there is no change in the yearly concrete volume, composition, and applications. The production of cement and concrete of the year 2015 is taken as basis for the calculations. The CO_2 uptake in the use stage is 365 000 tonne. The CO_2 uptake in the "recycling stage" (end-of-life stage) is 83 000 tonne.

The "CO₂ emissions due to cement consumption" is 1 960 000 tonne. It is not quite clear but it can be derived from the reasoning below, that this figure includes both CO₂ from calcination and fuel combustion. The cement consumption is 4 000 000 tonne and the average CO₂ emission per tonne cement is reported to be 0.49 tonne/tonne, 4 000 000×0.49=1 960 000 tonne. The slag content is said to be 36 %. If the rest is assumed to be clinker, we have a clinker consumption of $0.64\times4~000~000 = 2~560~000$ tonne (and a slag consumption of $0.36\times4~000~000 = 1~440~000$ tonne). The corresponding calcination emission can be estimated to $0.515\times2~560~000 = 1~318~000$ tonne CO₂. This leaves only 1~960~000 - 1~318~000 = 642~000 tonne CO₂ for the fuel part, which can be reasonable since CO₂ neutral fuels are used to a large extent².

² Personal communication with Edwin Vermeule: A lot of secondary fuels are used in the Netherlands (> 80 %). Those fuels are, to a large extent, CO₂-neutral.



Comparison with Simplified methodology Tier 1A

The Simplified methodology gives an uptake of $0.20 \times 1318000 = 263600$ tonne, plus the uptake in the GGBS $0.36 \times 4000000 \times 35 \times 0.001 = 50400$ tonne.

In total 314 000 tonne, corresponding to 86 % of the Advanced value.

Comparison with Simplified methodology Tier 1B

The Simplified methodology gives an uptake of $0.15 \times 1318000 = 198000$ tonne, plus the uptake in the GGBS $0.36 \times 4000000 \times 25 \times 0.001 = 36000$ tonne.

In total 234 000 tonne, corresponding to 64 % of the Advanced value.

The Irish method [5]

The carbonation rate for each application is calculated using a calibrated Irish formula from Silva et al. [15]. The degree of carbonation is not specified. For the year 1972, the calcination emissions are estimated to have been 780 000 tonne and the concrete produced that year is estimated to have taken up 98 000 tonne by the end of year 2013 (40 years), corresponding to 13 %. On a 100 years perspective, this should have increased to about $13\sqrt{(100/40)} = 20$ %. The report says 16 %. The figure 16 % is used in the statistical calculation for the Simplified methodology. It also mentioned that this figure corresponds to an uptake of 75 kg CO₂/tonne cement over 100 year service life. With a mean cement consumption of about 2 000 ktonne, it means an annual uptake in Ireland of about 150 000 tonne.

An estimate of the cement use in Ireland over last 40 years has been done, but it is not clear if these historical year-by-year values have really been used in the calculations. The uptake during one year in the existing buildings is not explicitly reported, but if it anticipated that it is approximately equal to 100 years of uptake in one year production, the value of 16 %, or 75 kg/tonne cement, can be used.

Comparison with Simplified methodology Tier 1A

The Simplified methodology gives the uptake of 0.20 or 20 %, which in this case is more than the presented Advanced method value of 16 % (20/16 = 125 %).

Comparison with Simplified methodology Tier 1B

The Simplified methodology gives the uptake of 0.15 or 15 %, which in this case is very close (15/16 = 94 %) to the presented Advanced method value of 16 %.

The Swiss method [13, 12].

The degree of carbonation is in the report [13] set to 75 % throughout, but is later in another paper [12] changed to 50 %. As in most reports, it is anticipated that the uptake during one year in existing buildings is equal to the uptake during (100) years in the concrete produced during one year. An average uptake of 10 % of the total emissions, for 50 years is presented in the report. For the calcination emissions, the figure can be estimated to be $10\times(1/0.63) = 16$ %. For 100 years, the figure can be estimated to be $16\times\sqrt{(100/50)} = 23$ %. With a degree of carbonation being 50 % instead of 75 % we get $23\times(50/75) = 15$ %.

(According to the Annex BB of the EN 16757 [16] the degree of carbonation varies between 40 % for indoor structures to 85 % for outdoor structures not sheltered from rain.)

Comparison with Simplified methodology, Tier 1A

The figure 15% is less than the Simplified methodology, Tier 1A, value, which is 20% (20/15=133%).



Comparison with Simplified methodology, Tier 1B

The figure 15 % is the same as the Simplified methodology, Tier 1B, value (15/15 = 100 %).

The Spanish method [6,7,8, and 9]

The Spanish results are extensively discussed and dealt with under Chapter 5.2.3, resulting in the Spanish CO₂ uptake value being 15 % of the calcination emissions.

Comparison with Simplified methodology, Tier 1A

The figure 15% is less than the Simplified methodology, Tier 1A, value, which is 20% (20/15=133%).

Comparison with Simplified methodology, Tier 1B

The figure 15 % is the same as the Simplified methodology, Tier 1B, value (15/15 = 100 %).

(This assumes that the Spanish value for comparison is the one updated by the C. Andrade paper [9] and the further calculation in Chapter 5.2.3.)

5.3.3.2.1 The Onward calculation method

Four of the national methods estimate the annual uptake in the concrete product stock by using the assumption that the uptake during one year in concrete that was produced during the previous (100) years, (which is the desired information) can be placed on an equality with the uptake the coming (100) years in concrete produced during the same year, (Advanced methodology II).

The uptake in existing concrete product stock during a certain year (the reported year) is what is to be calculated. This can be done by using knowledge of the historical concrete production (100 years if possible) as in Advanced methodology I. This knowledge is however often not at hand and the uptake calculation is also laborious. As an alternative, one can instead make use of the assumption that this "true" uptake value can be equivalent with the uptake the coming (100) years in concrete produced the same year (the reported year). This equality can be shown to be exact valid during certain circumstances and approximate valid during practical situations.

For an exact equality, it is required that the cement or rather concrete production has been almost equal the previous years and the actual year (the reported year). This means not only the amount and type of concrete but also the use of the concrete, i.e. the applications with their exposure and surface to volume ratio. See further explanation below *).

In practise, some circumstances modify these stringent requirements. Due to the square root of time dependency of the carbonation rate, it follows that the remote years production is of less importance for the present uptake. Even if we count on 100 years of service life for concrete contributing to the CO_2 uptake, the last 10 years of production of this period accounts for $\sqrt{10}/\sqrt{100}$ = 32 % of the total uptake. 20 years of production accounts for $\sqrt{20}/\sqrt{100}$ =45 %. This means that the requirement on similar production of concrete and applications can be limited to the last 10 or 20 years. However, if the cement use is very different in previous years (e.g. 50 - 20 years ago), then it is recommended to use the Advanced methodology I in Tier 3.

To improve the approximation, one should use the average value of concrete production during the last years (for instance estimated through cement consumption, which is generally well reported) for the onward calculation of the uptake.



The relative changes in cement distribution for different applications, due for instance to market reasons, seem to even out in the long run. It is thus normally more important to have a detailed knowledge of the cement distribution on different applications and products and of their exposure and surface/volume ratio on one occasion, than a vague picture at many occasions.

These facts accordingly make the approximation of the onward calculation methodology (Advanced Methodology II) to be an acceptable calculation method.

*) This follows from the equation for the CO₂ uptake.

 CO_2 -uptake = $(\Sigma(k_i \times DOC_i \times A_i)) (\sqrt{(t)/1000}) \times U_{tcc} \times C$

where all the parameters (k, DOC, A, Utca and C) are kept constant, their product being say P.

The uptake for the reported year, according to Advanced Methodology II, will, for 100 years, be $P \times \sqrt{100}$

The uptake the reported year, according to Advanced Methodology I, will for 100 years of service life be:

 $P \times \Sigma(\sqrt{i} - \sqrt{(i-1)})$, where i goes from 100 to 1. The result is also $P \times \sqrt{100}$.

(The uptake for the reported year is the difference between this year and the year before, summarized for all the previous years: $(\sqrt{100} - \sqrt{99}) + (\sqrt{99} - \sqrt{98}) + ... + (\sqrt{2} - \sqrt{1}) + (\sqrt{1} - \sqrt{0}) = \sqrt{100}$.)

5.3.4 Discussion of "good practice" for the Advanced methodology in the Use stage

The Advanced methodology should be based on the uptake in the actual existing concrete product stock. To achieve this, we first need the historical cement use in a country or region in question. This is often not a problem, since the cement production and use are normally recorded. Moreover, it might be enough to go 10 years back, since the uptake is following a square root of time dependency. Even if we can count on 100 years of service life for concrete contributing to the CO₂-uptake, the last 10 years production of this period accounts for $\sqrt{10}/\sqrt{100} = 32$ % of the total uptake the present year. The last 50 years production accounts for $\sqrt{50}/\sqrt{100} = 71$ %.

Thus, what happened in the cement consumption and concrete production a long time ago is of less importance for the present uptake. Also smaller production figures, in the remote years, often make this period less important. However, when there is available statistics on cement consumption for a longer period, a more accurate estimation can be done, as in the Swedish method [2].

As we have seen in the survey above, many countries rely only on one-year figures. The difference to the Simplified methodology lies in the necessary knowledge about concrete applications, of cement distribution on different applications/products. That is their relative amount, exposure, mean surface to volume ratio, concrete quality, cement content, k-values and degree of carbonation. The rate of carbonation for each application is calculated using k-values and the square root of time dependency. The k-values given in the EN 16757, Annex BB [16] for different exposures and concrete qualities have been found to be well balanced in many studies. The important factor degree of carbonation (DOC) can also be found in the EN 16757, Annex BB for different exposures or in Table 2.

The comprehensive earlier studies by Spain [6,7,8], compiled and adjusted in the recent paper by C. Andrade [9], includes many different types of cement, two types of concrete and three exposures,



compiled for the IPCC application, show that the DOC for indoor exposure is around 45 % and for outdoor exposure around 60 % for sheltered concrete and 80 % for not sheltered concrete (exposed to rain). (DOC is here expressed as the amount of CaO that has converted to CaCO₃ in relation to the original amount of CaO, or (which is the same) the amount of bound CO_2 in relation to the emitted CO_2 by calcination.) The Spanish figures, valid for Portland cement, are rather well in line with the ones presented in the EN 16757. The indoor is a bit higher (45 % compared to 40 %) and the outdoor is a bit lower (60 % or 80 % compared to 75 or 85 %).

Below is a first attempt to describe use and requirements for the two advanced methodologies.

5.3.5 The Advanced methodology I, based on wide historical data on cement and concrete use

Calculation method using historical data, according to, for instance, the Swedish and the Global studies [2,4], where the uptake each year is calculated as the difference between two consecutive years and is summed up for all the applications/products. Data from at least 50 years statistic of cement consumption (cement production-export+import) and knowledge of concrete use from at least three different and well-distributed years over the calculation period should be used. This methodology is proposed to be used for "Tier 3" described in Chapter 5.4.

5.3.6 The Advanced methodology II, based on a single or few years data on cement and concrete use

The Advanced methodology II is proposed to be used for calculations at Tier 2. The mean value of at least 10 years of cement consumption (all available data shall normally be used) may be used together with knowledge of one year of concrete use. The mean value of the cement consumption should normally be easy to access and would give a better estimation of the amount of carbonating structures than only the last year's consumption. It is anticipated that the uptake during one year in existing buildings is equal to the uptake during (100) years in the concrete produced during the same year. Calculation method onwards, according to for instance the Norwegian study [3].

Both methods (AM I and II) require knowledge of cement consumption statistics and concrete production and use, but on a different level. The calculation as such is simpler with the second methodology. The result should be presented as annual amount of CO₂ uptake for a country or region.

A suggested general description of the required steps in the Advanced methodology II is presented below, based on parts of the Swedish study above and the article by C. Andrade [9].

➤ Identify typical concrete applications (frequent type of structures). Normally, at least 5 applications should be needed, corresponding to at least 65 % of the cement consumption. For instance: Bridges, residential buildings, office buildings, roof tiles, pavement, shotcrete, sleepers, and mortar. Applications outside the chosen ones are treated as the most similar of the defined ones.



Cement content, concrete quality, exposure and specific surface (m² surface/m³ concrete) is described for each application. Especially exposure and specific surface may vary within the application.

➤ Calculate the CO₂ uptake per m³ for each application as a function of time:

The CO₂ uptake in kg per m³ concrete for each application during t years can be calculated as: (The formula can be found in EN 16757, Annex BB.)

 CO_2 -uptake = $(\Sigma(k_i \times DOC_i \times A_i)) (\sqrt{(t)/1000}) \times U_{tcc} \times C$

Where

CO₂-uptake is the total CO₂ uptake in kg CO₂/m³ concrete.

 \mathbf{k}_i is factor for the rate of carbonation for surface i in mm/ $\sqrt{(t)}$.

DOC_i is the degree of carbonation for surface i.

 A_i is the area of surface i in m^2 .

t is the number of years.

 U_{tcc} is the maximum theoretical uptake in kg CO_2/kg cement. The value is ≈ 0.49 for Portland cement (CEM I).

C is cement content in kg cement/m³ of concrete.

(It is also possible to calculate an application with different U_{tcc} or C, by giving them index i and put them in the Σ – parenthesis.)

Values of k can be found in EN 16757, Annex BB. See also Table 2.

Values of DOC can be found in EN 16757, Annex BB. See also Table 2.

Examples of this kind of calculation can be found in EN 16757, Annex BB, example 5 and 6.

- ➤ Calculate the last 20 years mean annual cement consumption and estimate the annual concrete production and the distribution on the different applications.
- ➤ Calculate the CO₂ uptake for each application during 100 years. (normally)
- ➤ Calculate the sum of the annual CO₂ uptake of all concrete applications.
- ➤ The sum of the cement content in the produced concrete should always be checked against the cement production.

5.3.7 End-of-life stage and secondary use

5.3.7.1 End-of-life stage (demolishing, crushing and storage)

In the Simplified methodology, we introduced a conservative "default" value for this phase based on the present normal handling procedures of demolished and crushed concrete. These procedures include normally storage in large unsheltered piles, during a rather short period of time.

Moreover, the recycling rate (the annual amount of demolished and crushed concrete in relation to the annual production) is normally low in most countries. The volumes can however be anticipated to increase in the future, since more concrete structures reach the end of the service life. This foreseen increase should however not be speeded up for CO₂ uptake reasons. The present normal



handling procedures for storage of crushed material, awaiting secondary use, are however not focused on facilitating CO₂ uptake and here; improvements can and should be implemented.

The drawback with the normal handling is the large piles of crushed material of mixed sized fractions. Although the fine fractions carbonate rapidly near the surface of the stockpile, the mixed size fractions, exposed to rain, form a rather compact material that prevents the free air circulation into the bulk of the stockpile. Measurements have shown that only up to about 0.3 m from the surface is influenced by carbonation. Changing of the handling procedures comprise commercial considerations, but can be very effective when it comes to CO₂ uptake.

In [17], there is an estimate that an enhanced processing of crushed concrete, implying 1-4 months storing in five fractions, can increase the uptake in the end-of-life stage to about 20 kg $\rm CO_2/m^3$ of concrete.

In the report [14], crushed concrete aggregates from waste hollow core slabs and decommissioned railway sleepers were analysed for their material characteristics. To investigate if graded crushed materials would have a better ability to absorb CO₂, the two different concrete materials used in this study were sieved and placed outdoors and sheltered from rain. The concrete was crushed and graded into three sizes; 0-4, 4-8, and 8-16 mm and placed in pallet collars with a net at the bottom. The total depth of the material was 0.4 m. After 18 months, the material was tested at four levels; 0, 0.12, 0.24, and 0.36 m from the top surface. Both the degree of carbonation and the amount of cement paste in each fraction was measured.

It was found that the grading of the crushed concrete aggregates into the three size fractions improved the CO_2 diffusion into the cement paste and increased the CO_2 uptake in the aggregates compared to the uptake of ungraded material. The main portion of the cement paste accumulated in the 0-4 mm size fraction and this fraction also experienced the highest carbonation degree; 0.45 - 0.65, but only in the outer layer. The inner layers carbonated at a lower rate, carbonation degree about 0.20, probably due to the higher gas diffusion resistance created by the finer particles. In the size fractions 4-8 and 8-16 mm, less gas diffusion resistance was experienced and the carbonation degree was similar throughout the whole material.

From the given data on amount of cement paste and degree of carbonation of the different size fractions, it is possible to calculate the CO₂ uptake with the used storage conditions (18 months outdoors, sheltered from rain).

The cumulative grading curve of all the crushed material was not measured at the investigation but can be taken to be similar to a curve of crushed concrete presented in [17]. The following relative figures for the three fraction sizes are anticipated.

- 0 – 4 mm: 50 % - 4 – 8 mm: 15 % - 8 - 16 mm 35 %

In the 0-4 fraction, the "outer layer" constitute about 25 % (carbonation degree 0.55), and the rest about 75 % (carbonation degree 0.20). The uptake per m^3 of concrete, crushed into the three fractions, can be calculated as shown in Table 4. The total reported amount of paste in the three fractions is 378 kg/ m^3 , where average quality concrete has original cement paste content of about 500 kg/ m^3 . About 24 % (122/500) of the cement paste has been lost during the crushing and fractioning process, probably as fine dust.



Table 4	Calculated CO ₂ uptake after 18 months (outdoors, sheltered from rain) in crushed concrete (1 m ³)
separated	l into fractions, [14].

Size fraction	Relative	Paste		Cement*	Degree of	CO ₂ uptake**
(mm)	amount	(%)	(kg)	(kg)	carbonation (%)	$(kg CO_2/m^3)$
						concrete)
0 - 4 outer	0.5×0.25	25	75	53	55	14
0 - 4 inner	0.5×0.25	25	225	158	20	15
4 - 8	0.15	10	36	25	25	4
8 - 6	0.35	5	42	29	30	4
Total	1.0		378	265		37

^{*} Paste, kg × 350/500. Cement 350 kg/m³, Water 150 kg/m³, w/c = 0.43 assumed (70 % of paste)

From Table 4, the CO_2 uptake for the three fractions after 18 months is 37 kg/m³ concrete. If one assumes that the uptake is proportional to the square root of time, this corresponds to about 18 kg/m³ concrete for 4 months storage.

Some part of the hardened cement paste will separate during the crushing process as very fine particles. With suitable handling, these very fine particles will rapidly carbonate, possibly in days but up to a few weeks. If about 20 % of the cement paste is falling in this very fine fraction, and the carbonation degree is 75 %, the resulting uptake for a concrete with CEM I-cement content of 300 kg/m^3 will be an extra $0.2 \times 300 \times 0.49 \times 0.75 = 22 \text{ kg CO}_2/\text{m}^3$ of concrete.

(The amount of these very fine particles is depending on crushing and sieving equipment and not yet possible to estimate accurately.)

It is important to find some way of including a future larger uptake in the end-of-life stage and to give reporting opportunities for countries with an already high rate of recycling as for instance the Netherlands and Germany. For very long storage times, years or more, and small particles in fractionized piles of low thickness, an uptake of about 75 % ([11], [16]) of the maximum theoretical potential uptake can be expected, corresponding to about $110 \text{ kg CO}_2/\text{m}^3$ of concrete.

The following estimations for CO₂ uptake in the end-of-life stage can be given.

For normal handling procedure or recycling rate less than 5 %:

(Same as for Simplified methodology, Tier 1)

If the annual amount of concrete in the end-of-life stage is known (m³ per year), the uptake can be calculated as 10 kg CO₂/m³ of concrete.

If the annual amount of concrete in the end-of-life stage is not known, the uptake can be calculated as 2 % of the annual calcination emissions, (factor 0.02).

For improved handling procedure:

A preliminary suggestion is that the uptake could be set to 20 kg CO₂/m³ of concrete if an enhanced procedure with air access in the fractions and at least 4 months storage in at least three fractions is applied. In this case, the amount of concrete needs to be known.

The handling procedure for demolishing, crushing and storage, presented in [17], is an example of such improved procedure.

^{**} Cement × Degree of carbonation × 0.49



5.3.7.2 Secondary use

Only very small amounts of concrete structures that have reached their service life, actually ends up as waste (landfill), but are used in crushed form as, or in, a new product – secondary use. It is therefore important to base the uptake calculations on the real amounts of concrete to secondary use. The CO_2 uptake in secondary use is quite similar to the uptake in primary use, so similar calculation methods could be used. However, even more factors are unknown for secondary use, so it can be difficult to create general but accurate methods. The more exact methods are often quite specific and depend on the type of secondary use (country specific), and may be treated under Tier 3 in Chapter 5.4. Here, a more general method is proposed.

In the Simplified methodology, we introduced a conservative "default" value for this phase.

The annual amount of concrete presently being taken out of service and reused is, accordingly in most countries, small compared to concrete produced. The volumes can be anticipated to increase in the future, since more concrete structures reach the end of the service life.

For crushed material in secondary use, the estimation of the uptake is rather diverse, depending on the different possible applications such as

- Road base, filling material and similar unbound applications (large potential for carbonation).
- Aggregate for new concrete (limited potential for carbonation).

Uptake in these applications is also taking place in already, to some extent, carbonated material. (During the primary use and the end-of-life stage.)

The literature is rare on this subject.

Calculation of scenarios with different premises can give an estimation of reasonable uptake figures for these applications. To do this, we need however information about average carbonation status of the used crushed material, as well as carbonation rate of the new applications. This information is, so far, not available.

In [17], a theoretical discussion of a possible uptake in the secondary use is presented:

"The use phase of the crushed material is calculated to a practical CO_2 maximum uptake. This uptake is thus reached in different time periods indicated in the comments. The use applications assume a relatively free access to air (CO_2) . For the fractions larger than 4 mm, this can probably be achieved by using the material as filling materials in different construction applications and leaving openings in the aggregate construction for air circulation. The smaller fractions (0-4 mm) have a relatively compact structure due a large share of very fine particles. This indicates that the material should be used in thin structures. Examples of this can be top surface layers or slip control on roads. The applications for high uptake of CO_2 are relatively new and further development work is required. The CO_2 uptake is estimated for each application based on aggregate size and type of application."

"For large aggregates, it is important to keep in mind that CO_2 is only taken up by the cement paste and not the ballast materials. Larger aggregates can thus consist of a stone covered with cement paste. Usually the stone material is stronger than the cement paste so the crushing fractures occur in the cement paste leaving a



stone with a relatively thin layer of cement paste. This means that the size distribution of the ballast used in the concrete can influence the CO_2 uptake. Thus, a relatively large aggregate can show a fast carbonation. The thickness of the cement paste layer is, in this case, of significant importance."

Also in [2], the large potential of uptake in secondary use applications is emphasized.

As a lowest level for secondary use, an uptake according to the Simplified method (Tier 1) can be used.

That is:

If the annual amount of concrete entering the secondary use is known (m^3 per year), the uptake can be calculated as 10 kg CO_2/m^3 of concrete. This value is valid for unbound applications of crushed material. (If the annual amount of concrete entering the secondary use is not known, the uptake can be calculated as 1 % of the annual calcination emissions, factor 0.01.)

Under favourable conditions for the secondary use applications, the <u>total uptake (primary use + end-of-life + secondary use)</u> can amount to about 75 % ([11], [16]) of the maximum theoretical potential (equal to the calcination emission), corresponding to about 110 kg CO_2/m^3 for an average concrete.

5.3.8 Cement with different constituents or additions added at the concrete mixer

Different cement constituents and concrete additions are frequently used in different parts of the world, but the degree of use varies considerably between different manufacturers and countries. The materials used are also very different and their ability to bind CO₂ is also different. The additions that are mainly used are ground granulated blast-furnace slag (GGBS), fly ash, silica fume, and limestone. Relatively few data are available for the uptake of CO₂ in different additions, but some data, especially for blast-furnace slag, exist. It is therefore difficult to provide general recommendations for calculations of CO₂ uptake in addition, why the recommendation is to include additions only when calculations are made using the Tier 3 method in Chapter 5.4. Therefore, the Tier 3 method should also be used when the use of additions is high in used concrete and if this affects the CO₂ uptake to a significant extent. However, to improve and simplify the uptake calculations, the CO₂ uptake due to use of ground granulated blast-furnace slag (GGBS) in concrete is included also in the proposed calculation method for Tier 2 in Chapter 5.3 and described in the present chapter. Below is a brief technical information about CO₂ uptake in additions that can be used for further work under Tier 3.

One has to distinguish between two effects of the additions. One is fairly well established and that is the increased rate of carbonation. Adjustment factors for the rate of carbonation compared to Portland cement (CEM I) is found for instance in the EN 16757, Annex BB [16]. For concrete of the same strength, the factors show for instance that the rate of carbonation will increase with 25 % if a binder with 40 - 60 % of GGBS is used. The rate will increase with 5 % if a binder with 10 - 20 % of fly ash is used. The increased rate of carbonation does however not mean that the total CO_2 uptake will increase. The uptake is also depending on how much of the $Ca(OH)_2$ and other compounds in the carbonated zone of the concrete that is actually converted to $CaCO_3$, (so called degree of carbonation).



GGBS, having a rather high CaO content, is the addition with the largest potential to take up CO_2 . It is a latent hydraulic binder and does not need to consume $Ca(OH)_2$ from the clinker hydration to react. The fly ash and especially the silica fume contain small amounts of CaO and need the $Ca(OH)_2$ from the clinker hydration to react. Since $Ca(OH)_2$ is normally the first component of the hydrated cement to carbonate, the addition of fly ash or silica fume, can theoretically decrease the amount of CO_2 uptake, even if the rate of carbonation will increase as mentioned above.

The limestone addition is normally used in amounts less than 20 %. The increase on rate of carbonation from that addition is 10 % or less. Only minor parts of the limestone participate in the hydration, so the uptake can be assumed to take part only in the clinker reaction products.

To summarize: We have a situation with increased carbonation rate, meaning larger volume of carbonated concrete. On the other hand, a decreased amount of CO₂ bound per volume in this carbonated concrete. This fact makes it reasonable that, for the estimation of uptake in the Simplified method, (Tier 1), see Chapter 5.2.1, it is proposed that it is counted in the clinker only and the figure is based on the annual clinker calcination emission.

A solution with zero CO_2 uptake in the additions is also used in the EN 16757, Annex BB as a conservative estimation. An estimation based on clinker content only may, however, underestimate the CO_2 uptake if GGBS is used as additional constituent or addition. The problem is that there is, so far, not much knowledge of the CO_2 binding capacity of the GGBS hydration products. The attempts to estimations relate the uptake capacity to the amount of CaO, in the same way as with the clinker. The amount of CaO in GGBS is lower and varies more than in clinker.

In the **Norwegian report** [3], a calculation example is provided by which 70 % of the CaO in the GGBS can be carbonated. The amount of CaO in the GGBS is, in this case, 40 %. (The amount of CaO in the clinker is set to 65 %.) The additional uptake in the GGBS can be estimated to be $0.70\times0.40/0.65 = 0.43$ or 43 % of the clinker uptake. Thus, if the maximum uptake in the clinker is 515 kg/tonne, the maximum uptake in the GGBS is $0.43\times515 = 221$ kg CO₂/tonne. It should be emphasized that the Norwegian report does not claim that the figure 70 % for the CaO that can be carbonated is reliable. The binding capacity depends on the hydration products of the GGBS and to which extent they will carbonate under the normal partial pressure of CO₂.

We have earlier proposed that a conservative estimation is that 0.15 ($0.15 \times 515 = 77$ kg CO₂/tonne clinker) of the annual calcination emissions will be taken up, (or 0.15 of the maximum uptake). If we apply the same thinking on GGBS, then $0.15 \times 221 = 33$ kg CO₂/tonne GGBS will be taken up.

Note that for concretes with GGBS, the uptake per m³ of concrete is normally smaller than the uptake of concrete without GGBS, due to the fact that the clinker content is reduced and that GGBS take up less CO₂ per mass than clinker, (less CaO content and less "reactive" CaO).

In the **Dutch report** [20], it is anticipated that the maximum CO_2 uptake in GGBS is 140 kg CO_2 /tonne GGBS. The maximum uptake in CEM I is 490 kg CO_2 /tonne cement, or 515 kg CO_2 /tonne clinker. A maximum CO_2 uptake of 140 kg per tonne GGBS will result in an uptake of $0.15 \times 140 = 21$ kg CO_2 /tonne GGBS.

Without any more data than the values 221 and 140 kg/tonne for the maximum CO₂ uptake in the GGBS, or 33 and 21 kg/tonne for a conservative estimate of the uptake, it seems feasible to select a value of about 25 kg CO₂/tonne GGBS for the annual uptake in concrete with GGBS using the methodology in Tier 2.



The value 25 kg CO₂/tonne GGBS corresponds to the maximum emission value 166 kg CO₂/tonne GGBS, which emanates from the blast-furnace in the steel industry.

In line with the mean value factor 0.20 (see Chapter 5.2.3), the uptake in the GGBS can be estimated to $0.20 \times 166 \approx 35$ kg CO₂/tonne GGBS.

5.4 Tier 3 CO₂ uptake model - Advanced user developed models

5.4.1 Introduction and background

As previously stated in this report, the calculation of CO₂ uptake in concrete and other cement-containing products is associated with difficulties. In principle, it is not possible to direct measure the uptake of CO₂ on-site in existing concrete products. Only the carbonation depth can be measured and one is obliged to use calculation models to determine the CO₂ uptake. However by taking samples from the structures, CO₂ uptake can be determined in laboratory. A variety of factors also affect both the total uptake and the carbonation rate. For Tier 1 and Tier 2 in Chapters 5.2 and 5.3, simplified calculation methods have been developed to make approximate calculations of the CO₂ uptake. In order to make more accurate calculations of CO₂ uptake, more advanced computer models are needed that take into account and include many of the different factors that affect the CO₂ uptake. The uptake takes place especially in the concrete surfaces on the existing concrete products in society and it is therefore important to have a good knowledge of these surfaces. This allows calculation models to be based on historical data for cement and concrete use in each country, supplemented by estimates of uptake surface areas and their characteristics and conditions.

Research on carbonation of concrete has been going on for a long time and much background information for complex computer models are already available. As already stated in previous chapters, several countries have already developed complex computer models to make accurate calculations of CO₂ uptake in each country. In Tier 3 of this chapter, we want to open up the possibility for different countries to use complex national calculation models. Unlike the calculations in Tier 1 and Tier 2, where the designated calculation resources in each country are expected to be able to perform the calculations independently, the complex computer models require collaboration with various cement and concrete researchers. One can also imagine that computer models (software) for CO₂ uptake are developed at an international level, which can then be used in other countries following a national adaptation and with national input data. The complex computer models used may, of course, undergo a normal scientific review, which is a common procedure within the IPCC.



5.4.2 Proposed CO₂ uptake model for Tier 3

For Tier 3, the design of the software (computer model) can be determined by the respective provider. However, the software must maintain high quality and be built on a scientific basis taking into account the experience in the field both nationally and internationally. The software must be objective and neutral, and calculate the CO₂ uptake values as accurately as possible and with lowest possible uncertainty. Used methods and software should be transparent so that there is a possibility to conduct a scientific review of the entire system. The software must also be able to calculate the CO₂ uptake for a specific country or region, and it should also be specifically adapted to the respective geographical area. Furthermore, the software should also take into account as many as possible or most of the different aspects that affect the uptake of CO₂ in different concrete structures. Preferably, the software should also be based on the uptake of CO₂ in the actual calculated uptake areas based on historical data of annual amounts of clinker/cement or concrete used. Example of such methodology and models can be found in the peer reviewed articles [2] and [4]. Below is a list of aspects or data for CO₂ uptake in concrete or other cement-containing products that the software should be able to take into account and thus include in the model.

- ➤ Annual use of clinker/cement/concrete in the calculation region including historical data from the beginning of the time when cement began to be used. However, uptake data from primary products older than 80 years can normally be neglected as the CO₂ uptake is very slow. The use shall be calculated as (production-export+import) for the region.
- ➤ The use of additions such as blast-furnace slag, fly ash, silica fume, and limestone including also specific CO₂ uptake values for each material.
- ➤ If the use of additions in concrete is substantial and has a significant impact on the CO₂ uptake, the Tier 3 method shall always be used.
- > Emissions from the calcination of the materials in the cement kiln and check for the inclusion of emissions from additions for the corresponding year in the originating processes for the additions.
- ➤ Corresponding uptake areas for CO₂ in concrete structures or other cement-containing products, e.g. from the distribution in primary use for different concrete products and its area/volume ratio.
- ➤ Concrete surface environment when used in various products (porosity (w/c), temperature, indoors, outdoors, moisture, exposed to rain, underground, underwater, or surface treatment such as paint, wallpaper, asphalt/bitumen, etc.).
- Rate of carbonation including the square root of time dependence except for covered surface where a polynomial is more accurate.
- ➤ The sum of the cement content in the produced concrete should always be checked against the cement production.
- ➤ End-of-life processes for concrete or other uptake products during the calculation period including historical data. This shall include amount of concrete for demolition, type of handling (e.g. crushing, storing). How crushing into smaller fragments will increase the uptake surfaces and thereby the carbonation rate. Calculation of CO₂ uptake in the end-of-life process.
- ➤ Production of secondary products. Amount of concrete to secondary use products and its yearly CO₂ uptake, also for historical data.
- ➤ Landfill and use of remaining amounts and its CO₂ uptake. The CO₂ uptake in all end-of-life material must be included.



6 Discussion and conclusions

Climate aspects are now regarded as one of the most important environmental issues both globally and nationally. Yearly statistics on emissions of greenhouse gases (e.g. CO_2 , CH_4 and N_2O) have been collected in many countries during many years and the emissions have then been compiled at a global level. Emissions of greenhouse gases have an impact on a global level. Thus, where an emission occurs is therefore of secondary importance. From the outset there was a clear focus on CO_2 from fossil fuel combustion. These emissions remain a very important contributing factor to the greenhouse effect, but recent research has shown that the cause of the greenhouse effect is considerably more complex and several sources can contribute to the greenhouse effect. For example, CO_2 from combustion of biogenic fuels is not considered to contribute to the greenhouse effect when the growing biomass is considered to take up a corresponding amount of CO_2 , but this is only applicable if a replanting occurs. In the case of e.g. deforestation, in which no biomass regrowth occurs, CO_2 emissions from combustion of such biomass can be considered equivalent to CO_2 from fossil fuels, at least as long as the deforestation exists.

The insight into the complexity of climate and climatic effects has resulted in calculation methods for emissions of climate gases that have been developed and broadened so that they now contain several different types of greenhouse gas calculations. This research study includes such a widening of the climate gas calculations by taking into account the uptake of CO_2 in concrete through carbonation. Carbonation is a process that has been known for a long time in the concrete chemistry but now has a new scope for climate gas calculations. Since concrete is not chemically stable, the concrete reacts with CO_2 in the air under reformation of carbonates in the concrete. There is also a relationship between exhausted CO_2 from the materials (not the fuels) in the cement kiln and the amount of CO_2 that can be taken up by carbonation. However, the carbonated concrete is chemically stable so there is no risk that the CO_2 that has been taken up will return to the atmosphere. This can only happen if the concrete is heated to a temperature where calcination can occur, i.e. about > 850 °C.

Greenhouse gas calculations made within the framework of the IPCC and UNFCCC are usually different types of emission calculations. Emissions are often relatively easy to calculate and control measurements can usually be made, even though the amount of measurements that would have to be made can complicate the practical implementation. Simple linear calculations can often be used, using different emission factors. As regards the uptake of CO₂ through carbonation, the results of the present study show significant difficulties. Calculation of a country's annual uptake of carbon dioxide by carbonation of concrete is done by estimating the uptake in concrete surfaces on the concrete products in the country. The uptake in a concrete surface cannot be directly measured on site, but must be calculated based on laboratory and field measurements from previous research. In addition, the uptake is influenced by various factors such as weather (rain and moisture), surface coatings, surfaces under water and soil, and the quality of concrete. This increases the difficulty of developing simplified but necessary methods. In this study, an attempt has been made to develop such calculation methods (Tier 1 and Tier 2). However, experience from previous estimates in the field of research shows that more advanced calculation methods are to be preferred for this type of calculations. This study opens up such calculations in Tier 3. In order for such calculations to be implemented in most countries with high quality while being cost effective, common global computer models should be developed. These models can then be used for national calculations after a local adaptation to national inputs.



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Appendix 1 – Characterization of existing CO₂ uptake models

The project has developed an overview of the various national and international models available to calculate the uptake of CO₂ in concrete. These models usually represent more advanced and complex uptake models. Table 5 below presents the different models with references and summary information about each model for comparison.

Table 5 Overview and characterization of existing models for calculation of CO₂ uptake and their background.

	Sweden	EN 16757	Norway	Netherlands	Ireland	Switzerland	"Global"	Switzerland 2	Spain
General Framework									
Published	Andersson R. et al, "Calculating CO ₂ - Uptake for Existing Concrete Structures during and after Service Life", 2013	Product Category Rules for Concrete and Concrete Elements, 2017	Engelsen and Justnes, "CO ₂ binding by Concrete", 2014	Vermeulen E., Balans tussen emissie en opname CO ₂ , 2017	Fitzpatrick D. et al, "Sequestration of Carbon Dioxide by Concrete Infrastructure: A Preliminary investigation in Ireland", 2015	Nygard and Leemann, "Carbon diioxide uptake of reinforced concrete structures due to carbonation", 2012	Xi et al,"Substantial global uptake by cement carbonation",2016	Leemann, Hunkler, Widmer, "Calculation of CO ₂ -binding during service life of concrete", 2018	I. Galán et al, "Summary of the Study of the CO ₂ Sink Effect of Cement Based Materials" 2009 I. Galan et al, "Sequestration of CO ₂ by concrete carbonation" IECA Editors 2009
Number of references	23	23 in Annex BB on Carbonation	82	8	12	73	71	7	44
View	Country/society	Single product in future	Country/society	Country/society	Country/society	Product and country/society (Uptake in the total production during 50 years)	Global/society	Product (country/society)	
Perspective	Annual uptake	(PCR/EPD)	Annual uptake	Annual uptake	Annual uptake	, ,	Uptake 1930-2013		Uptake at 50 and 100 year
Base	Clinker consumption, historical 100 years	Clinker content in the product	Present cement consumption, 100 years future uptake	Cement consumption,	Clinker consumption, historical 40 years	Cement (clinker) consumption 2010	Cement consumption, historical 1930-2013, in four regions: China, US, Europe, rest of the world		Cement production
Constituents participating in the uptake	Uptake in clinker	Uptake in clinker	Uptake in CEM I, CEM II/A-V and CEM II/B-S	Uptake in CEM I (clinker) and in slag	Uptake in clinker	Uptake in clinker	Cement without specification. Cement kiln dust.	Uptake in CEM I, CEM II/A and CEM II/B	Clinker and fly ashes, limestone. Pozzolan and slags



	Sweden	EN 16757	Norway	Netherlands	Ireland	Switzerland	"Global"	Switzerland 2	Spain
FA and slag	Not taken into account	Taken into account in calculation of carbonation depth, but not for CO ₂ uptake	Taken into account	Slag taken into account for both carbonation depth and uptake (0.14 kg CO ₂ /kg slag)	Not taken into account	Not relevant	Only clinker content accounted	Not specified	YES included
Maximum uptake value	(Degree of carb)×(emission at calcination of clinker)	(Degree of carb) ×(emission at calcination of clinker)	(Degree of carb)×(emission at calcination of clinker + 70 % of uptake in CaO of slag)	(Degree of carb)×(emission at calcination of clinker + 0,14 kg CO ₂ /kg of slag)	(Degree of carb)×(emission at calcination of clinker)	(Degree of carb)×(emission at calcination of clinker)	(Degree of carb)×(emission at calcination of clinker)		Degree of carbonation by carbonated cement
Calcination emissions (kg CO ₂ /tonne cement)	CEM I 494 (CEM II 455)	Based on actual clinker content (CaO set to 65 % of clinker)	CEM I 471, CEM II/A-V 396, CEM II/B-S 332	CEM I 490, slag 0	CEM I 479, CEM II/A 428	508 kg/tonne clinker (CEM I 483, CEM II/A 406)	"Process" emissions (calcination) 1930- 2013 is 10.4 GtC	CEM I 474, CEM II/A 413, CEM II/B 357	Takes value of other works 550 kg/tonne clinker
Results									
Methods CO ₂ uptake	300 ktonne/year 2011, (125 kg/tonne cement)		140 ktonne/year 2011 (165 including recovery phase) (70 resp. 83 kg/tonne cement	365 ktonne/year 2015 (90 kg/tonne cement). Uptake in recycling stage 83 ktonne (21 kg/tonne cement) Adjustment 75 ktonne resp. 18 kg/tonne cement)	150 ktonne/year 2011 (estimated from the value 75 kg/tonne cement)	No absolute values presented, only relative.	Global uptake 2013 is 0.24 GtC. Uptake 1930-2013 is 4.5 Gt C.1 billion tons/year 2013 (250 kg/tonne cement)	No absolute values presented, only relative	Several quantities in function of type of concrete and exposure class
Calculated as	Uptake in existing building stock, estimated with the previous 100 years cement consumption		Uptake in the 2011 production of concrete during coming 100 years	Uptake in 2015 production of concrete during coming 60 years.	Uptake in one year production of concrete during coming 100 years	Uptake in one year production of concrete during coming 50 years	Uptake in existing building stock, estimated with the previous years (1930- 2013) cement consumption	Uptake in one year production of concrete during coming 100 years	Calculated as % of clinker fabrication INCLUDING THE FUEL (multiply by 1.7 the results for referring to decarbonation only
Corresponding to	17 % of production emissions year 2011		15 % of production emissions (18 % including recovery)	19 % of production (incl. import) emissions (23 % including end of life stage)	16 % of calcination emissions, corresponding to about 10 % of total.	Declared to be 10 % of the total emissions, corresponding to (1/0.63)×10 = 16 % of the calcination emissions.	Uptake 1930-2013 corresponds to 43 % of calcination emissions	17 % of calcination emissions	2.7 % of emitted CO ₂ by decarbonation
Sensitivity	Macro calculation can be overestimated a single year but not over time						Uncertainty analysis carried out.		Surface/volume ratio was calculated for several elements and the carbonation rates were experimental values



	Sweden	EN 16757	Norway	Netherlands	Ireland	Switzerland	"Global"	Switzerland 2	Spain
Accuracy	Known method and input data based on measurements	Known method and input data based on measurements	Known method and input data based on measurements	Known method and input data based on measurements	Conservative assumptions	Reduction with a factor 1.5 according to later paper by Leemann And Hunkler, "Carbonation of concrete: assessing the CO ₂ -uptake, 2016.	Known method and input data based on measurements		Conservative assumptions
Carbonation theory	x 11 1 1 1	Y 11 1 1	Y 11 1	x 11 1 1 1	0.7 1 (204.0)	x 11 1 1 1	x 11 1 1 1		** 1
Carbonation model references	Lagerblad and others. Nilsson	Lagerblad and others	Lagerblad and others. Nilsson (covered conc.)	Lagerblad and others	Silva et al (2014)	Lagerblad and others	Lagerblad and others		Various and own models
Depth of carbonation model	$d = k t^{\frac{1}{2}}$ Polynomial expressions for covered concrete. $d = f(t2, t, t^{\frac{1}{2}})$	d = k t ^½	$d = k t^{1/2}$ Polynomial expressions for covered concrete. $d = f(t2, t, t^{1/2})$	$d = k t^{\frac{1}{2}}$	Formula for RH less than 70 %, modified by factors 1.0 for internal, 0.5 external sheltered and 0.3 external exposed to rain.	d = k t ^½	d = k t ^½		$d = k t^{V_2}$
Exposure classes	11	6	6	5	Residential 6, Civil engineering 3, Commercial 7		5		Standard: interior, outdoors sheltered and non-sheltered from rain
Strength classes	4	4	4	7	5	3 (?)	4		2 concretes and paste
Carbonation degree	50-90 %, specified for each exposure	40-85 %, specified for each exposure	70 %	40-85 %, specified for each exposure	Not specified	75 %, changed to 50 % in the paper of Leemann and Hunkler, cf "Accuracy" above.	80 % for concrete 91.5 % for mortar		Main aim of the study
Concrete data									
No. of applications (type of product)	7	Single structures/prod ucts	24	5	3	20	Depending on region, most detailed from China		Concrete elements in general
Total no of product exposures (scenarios)	33	Infinite	54	25	14	31	Depending on region, most detailed from China		3
Application surface/volume	Calculated for each application	Actual structure	Calculated for each application	Calculated for each application	Calculated for each application	Calculated for each application	Calculated for each application		Calculated with detail
"special case"	overlapping corner						Mortar and cement kiln dust		No recycling, no secondary use or mortar application
Historical use of concrete in different applications, based on statistic	60 years (100 years with less accuracy)		Not applied (use 2011 provided)	Not applied (no change in consumption during the years is assumed, 2015 is the basis)	40 years	Not applied	Depending on region, US 1997-2005, China 1996-2012, Europe (nordic countries.) 2003.		Yes, several ages were studied



	Sweden	EN 16757	Norway	Netherlands	Ireland	Switzerland	"Global"	Switzerland 2	Spain
End of calculation	See maximum uptake value	See maximum uptake value	See maximum uptake value	See maximum uptake value	See maximum uptake value	50 years, see remark below			50 years for buildings and 100 years for infrastructure
End-of-life and secondary use									
End of life processes	Demolished, crushed, stockpiled and reused as unbound material	Demolished, crushed, and stockpiled. Reuse only as additional information	Demolished, crushed, stockpiled and reused as unbound material	Demolished, crushed, stockpiled, and reused.	Only shortly treated	Demolished, crushed, stockpiled, and reused.	Demolished, crushed, and secondary use. Only small uptake		No recycling or secondary use
Demolished and crushed, % of annual production	8 %		10 %	35 % (5.1 m ³ recycled/14 m ³ produced)		Scenarios of 0, 40, 60 and 100 % recycled			no
% recycled	Existing	National provisions	10 % of annual concrete production	35 %		Scenarios of 0, 40, 60 and 100 % recycled	China 3 %, USA 60 %, Europe 61 %, ROW 25 %		no
Stored to increase uptake	Potential given	Potential given							no
Remark	Uptake during one year(2011) in existing buildings is calculated from 100 years cement statistics and 60 years concrete application statistics		Uptake during one year in existing buildings is supposed to be equal to the uptake during 100 years in one year (2011) concrete production.	Uptake during one year in existing buildings is supposed to be equal to the uptake during 60 years in one year (2015) concrete production. Concrete blocks are responsible for a significant proportion (24 %) of the CO ₂ uptake.	Uptake during one year in existing buildings is supposed to be equal to the uptake during 100 years in one year concrete production. Open texture concrete blocks and roof tiles are responsible for a significant proportion of the CO ₂ uptake	The uptake figure 16 % of the calcination emissions should according to the later recommendations be reduced by a factor 1.5, that is to 10.7 %. In order to be consistent with the methods in the other countries a 100 years perspective should be applied, resulting in 1.414×10.7=15 %	Uptake during 1930- 2013 as well as present (2013). Uptake in mortar is significantly contributing to the large figures	This more recent but short report, present values for typical Swiss concrete house building applications	



Appendix 2 – Literature overview of carbonation

Uptake of CO₂ in concrete through carbonation is not a new research field. Carbonation has been known for a long time and an extensive research exists in the field. Originally, the research was conducted to understand how carbonation affects the properties of concrete. The formation of carbonates in the concrete makes concrete stronger, but reduces the pH of the concrete, which affects the corrosion properties of the reinforcement bars. Later research on carbonation of concrete has a stronger focus on analyzing the total CO₂ balance for cement and concrete. In Appendix 2, we want to show the research width that exists in the field as a complement and a basis for the model studies that have been carried out in this study.

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